

Toward an Understanding of GPCR-ligand Interactions

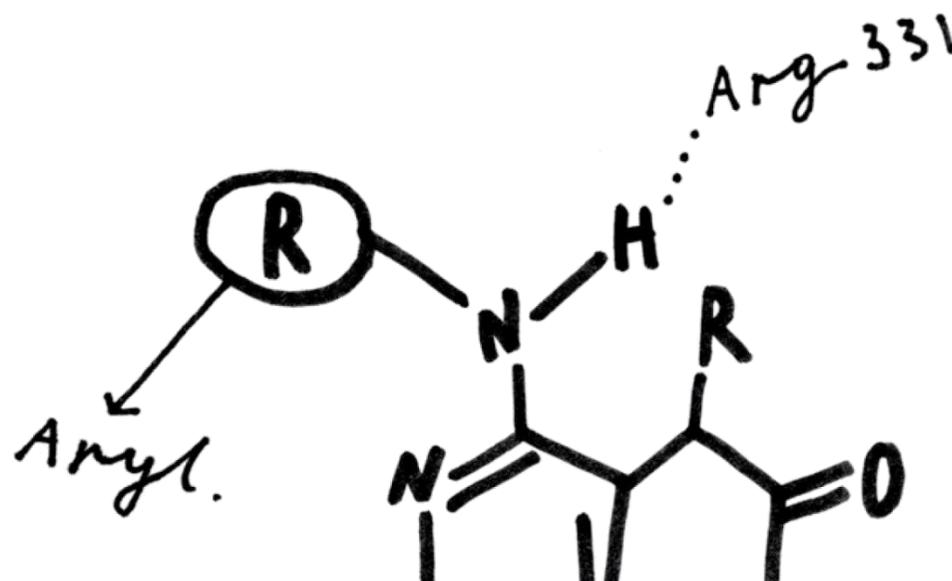
Alexander Heifetz



Agenda

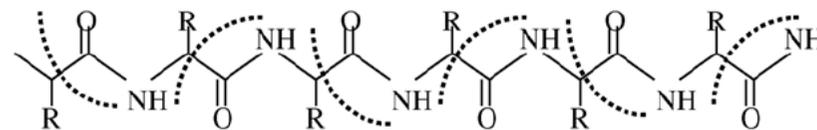
Fragment Molecular Orbitals (FMO) for GPCR exploration

HGMP-C4XD approach for GPCR drug discovery

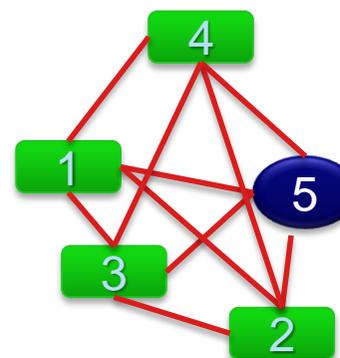


Fragment molecular orbital (FMO)

- “Visual inspection” and molecular mechanics cannot fully explain complex protein-ligand interactions
- FMO is a quantum mechanical method that has been developed¹ for application to large (biological) systems
 - takes 24h on 36 CPUs per protein-ligand system
- FMO provides detailed analysis of GPCR-ligand interactions and their chemical nature
 - Calculate individual contribution of each residue and water molecule to binding enthalpy
- Exploration of these receptor-ligand interactions provide key insights for further SBDD



Fragmentation of peptide



$$\Delta E_{IJ}^{\text{PIE/PCM}} = \Delta E_{IJ}^{\text{ES}} + \Delta E_{IJ}^{\text{EX}} + \Delta E_{IJ}^{\text{CT+mix}} + \Delta E_{IJ}^{\text{DI}} + \Delta E_{IJ}^{\text{SOLV}}$$



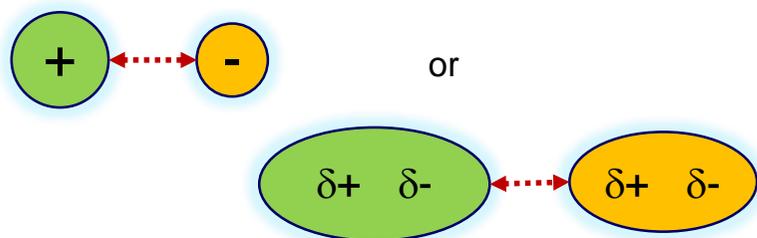
PIE (Pair Interaction Energy)

Pair interaction energy decomposition analysis (PIEDA)

Four energy terms obtained

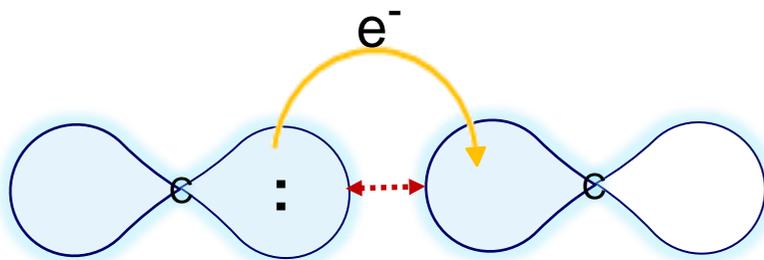
Electrostatic (PIE^{es})

Forces between point charges / (permanent and induced) dipoles/quadrupoles



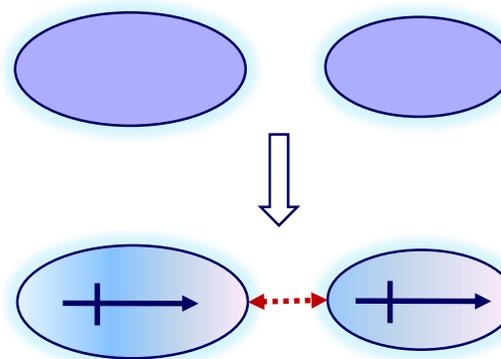
Charge transfer (PIE^{ct})

Interactions between occupied orbital of the donor and unoccupied orbital of the acceptor. Orbital energy gap and overlap are the important factors.



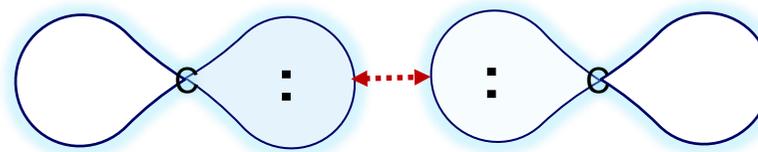
Dispersion (PIE^{dis})

Non-polar molecule can have dipolar moment for a short period of time due to the movement of electrons within the molecule; this induces weak dipole-dipole interaction



Exchange repulsion (PIE^{ex})

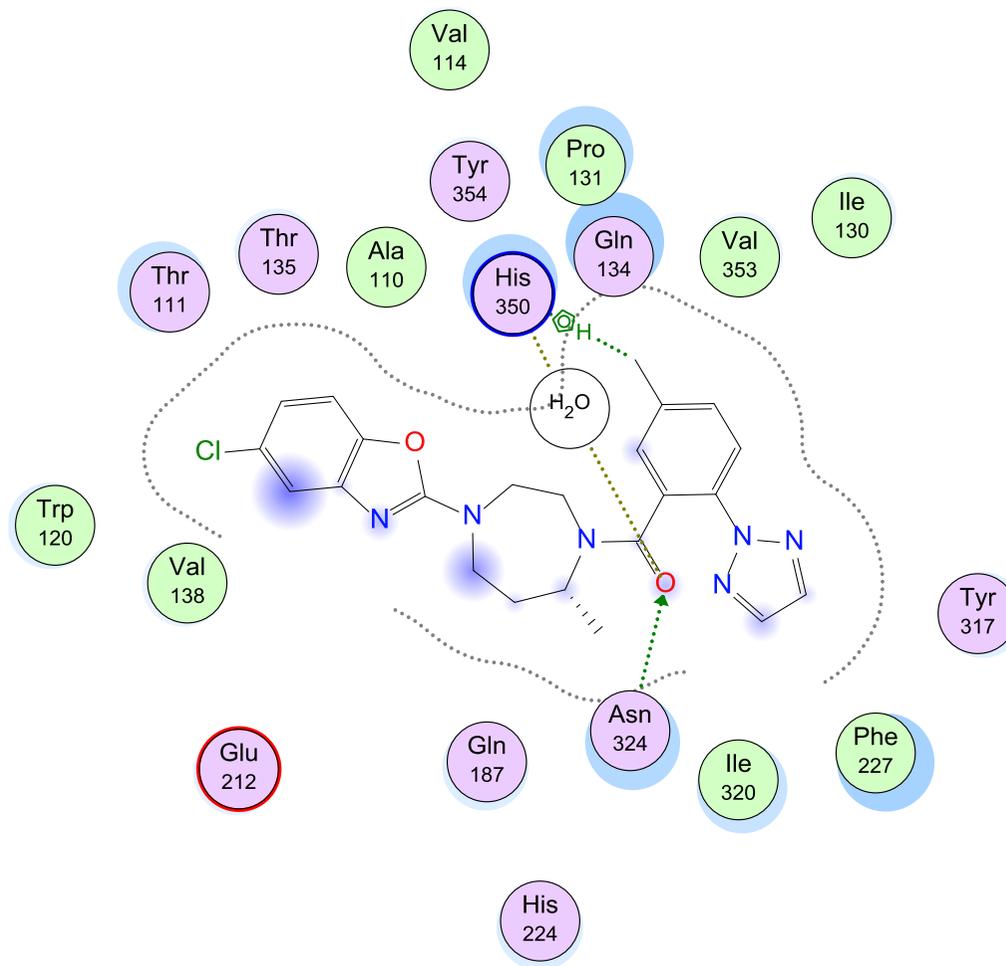
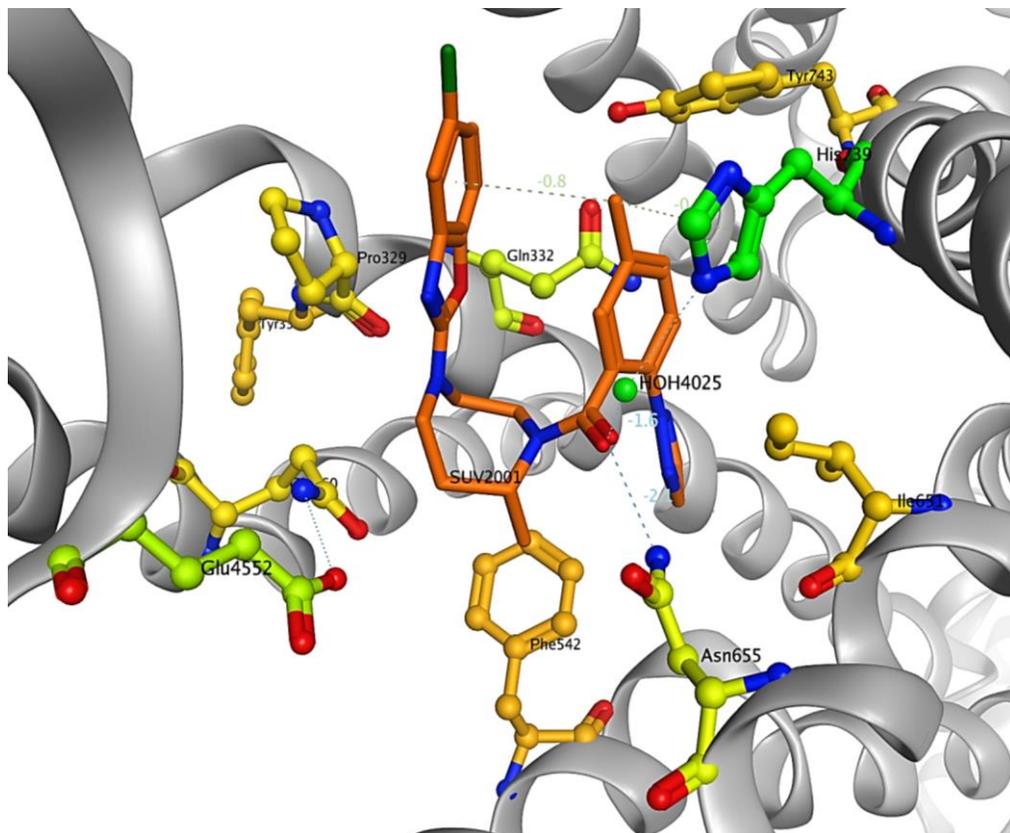
Forces between molecules placed close together and always repulsive. Mainly due to the overlap of two occupied orbitals.



FMO for hOX2-Suvorexant complex



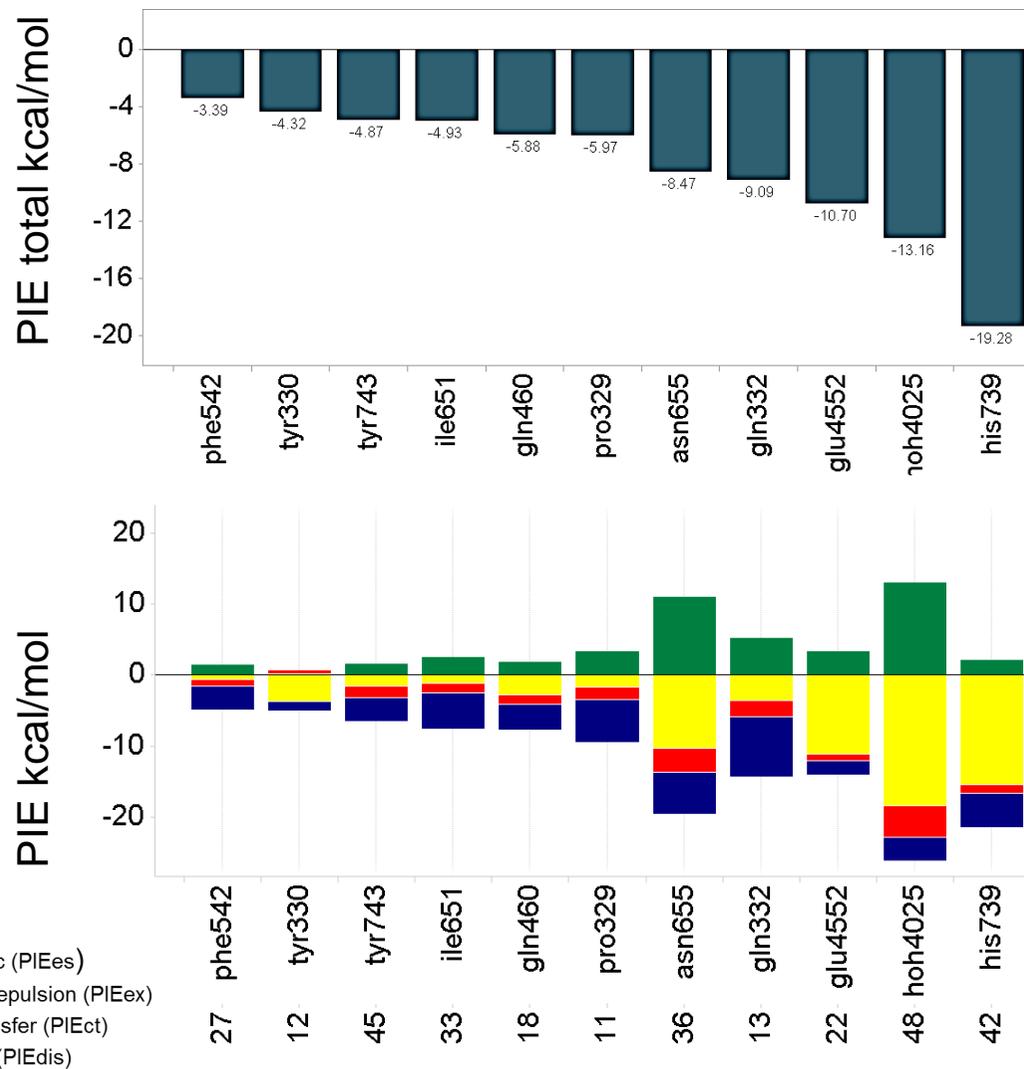
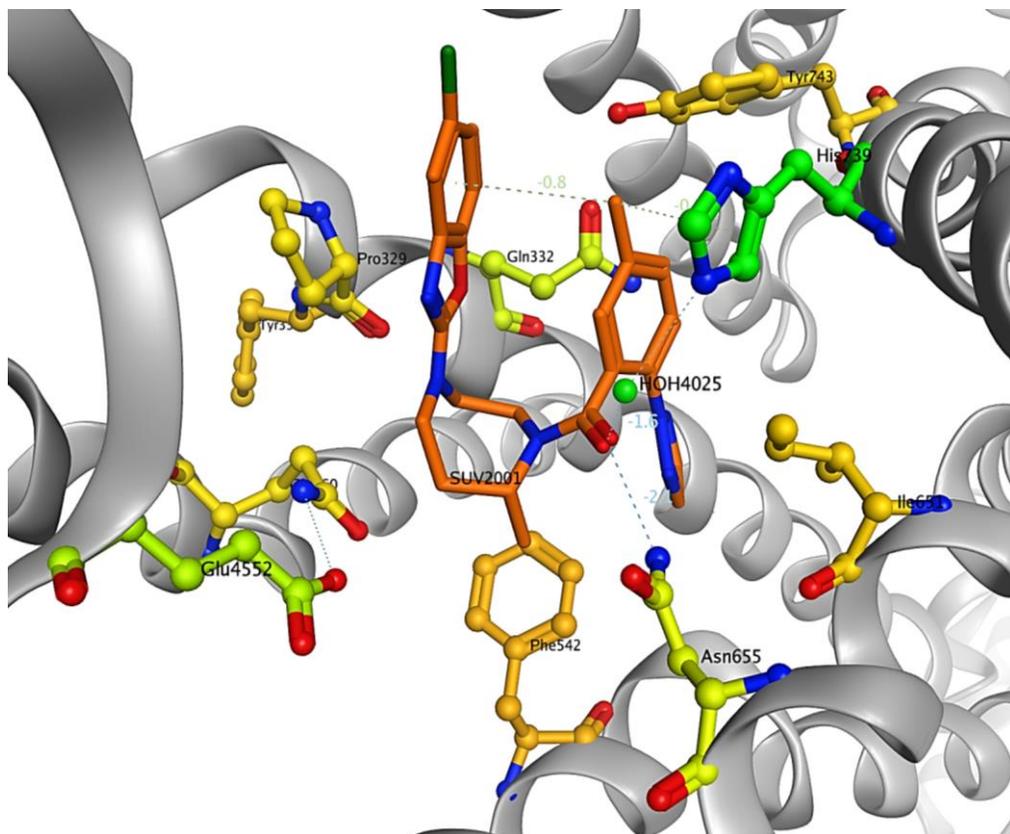
PDB entry 4SoV¹



- Suvorexant (hOX1R $K_i = 0.54$ nM, hOX2R $K_i = 0.35$ nM)²,

FMO for hOX2-Suvorexant

Filtered* PIE plots



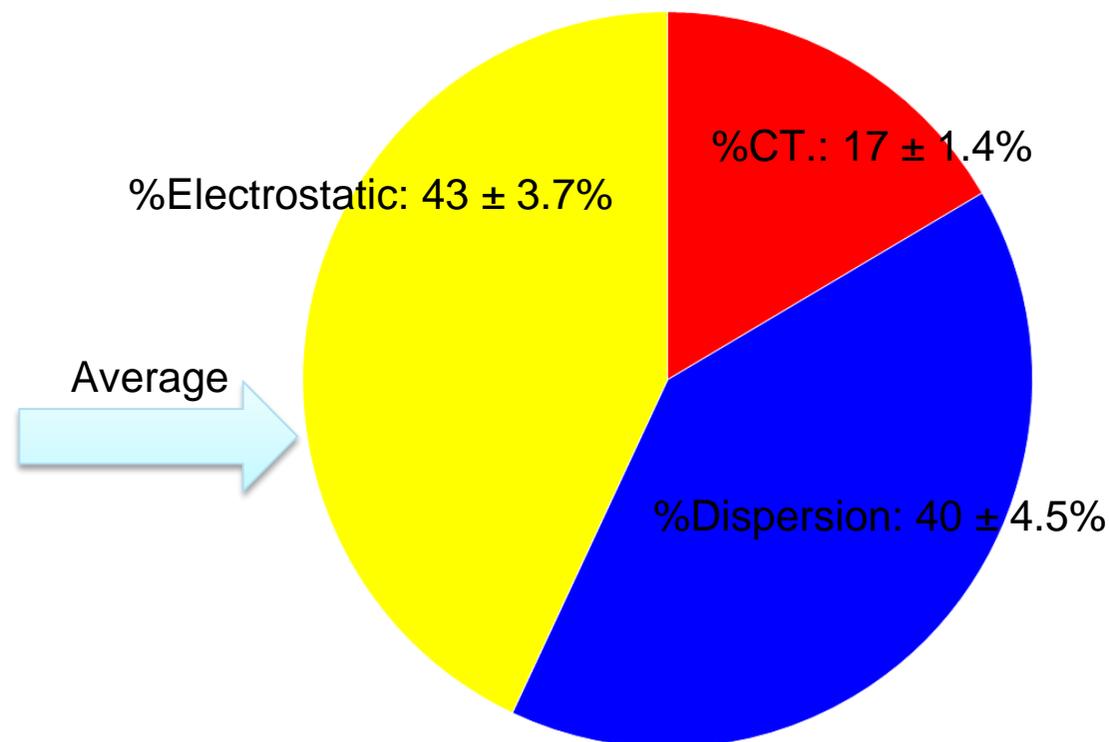
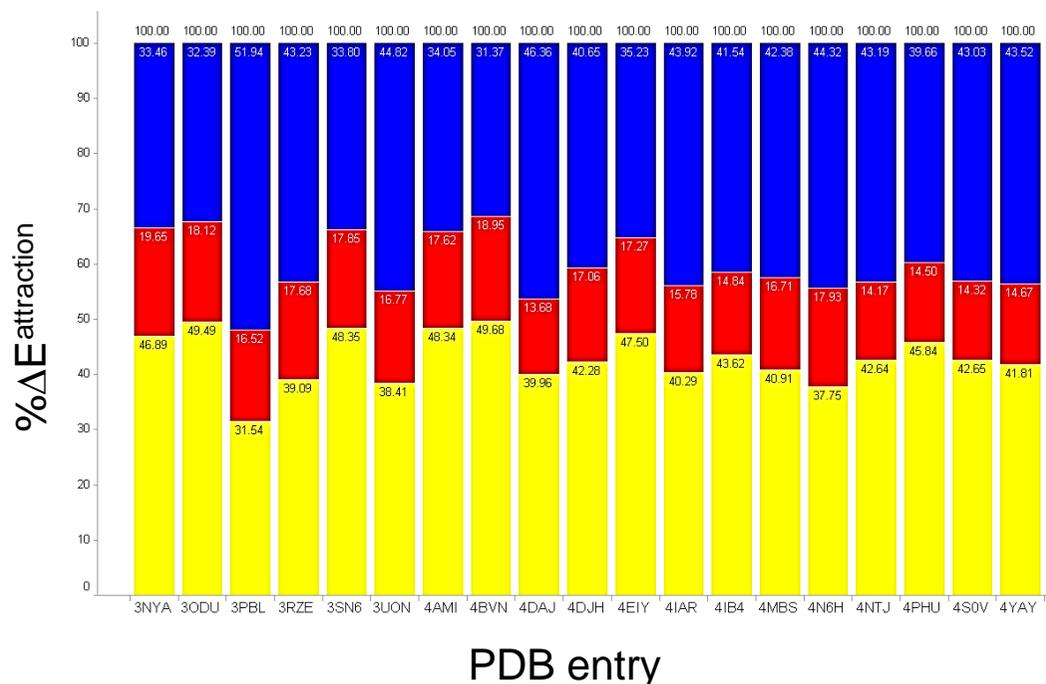
- Suvorexant (hOX1R $K_i = 0.54$ nM, hOX2R $K_i = 0.35$ nM, J Comput Aided Mol Des (2014) 28:5–12)

■ Electrostatic (PIEes)
■ Exchange repulsion (PIEex)
■ Charge transfer (PIEct)
■ Dispersion (PIEdis)

*FMO was calculated for all pocket residues in radius of $\leq 4.5\text{\AA}$ from the Suvorexant. In above plots only filtered PIE energies are showed \leq average PIEtotal (-2.0 kcal/mol)

Interaction signature

Analysis of 19 GPCR-ligand crystal structures



- Interesting observations:

- Hydrophobic interactions (Dispersion) has almost equal importance for ligand binding in GPCRs like Electrostatics
- Typical interaction proportion between %Electrostatic, %Dispersion and %CT is: 1:1:0.5 accordingly

$$\Delta E^{attraction} = \Delta E^{ES} + \Delta E^{CT+mix} + \Delta E^{DI}$$

FMO pros, cons and future plans

Pros:

- FMO provides important information on GPCR-ligand interactions and their chemical nature that cannot be explored using other non QM methods
- Structural insights provided by FMO are essential for further SBDD¹

Cons:

- FMO calculations requires accurate (good resolution) structures, which are not available for every GPCR or GPCR-ligand complex

Future plans

- To adapt the HGMP-FMO method² for optimisation of GPCR-ligand interaction geometries to a degree that will allow reliable FMO calculations
- To add solvation effect into FMO calculations by combining this method with the heterogeneous polarizable continuum model (het-PCM)³

¹Mazanetz M. et. al.; J Cheminform. 2011 Jan 10;3(1)

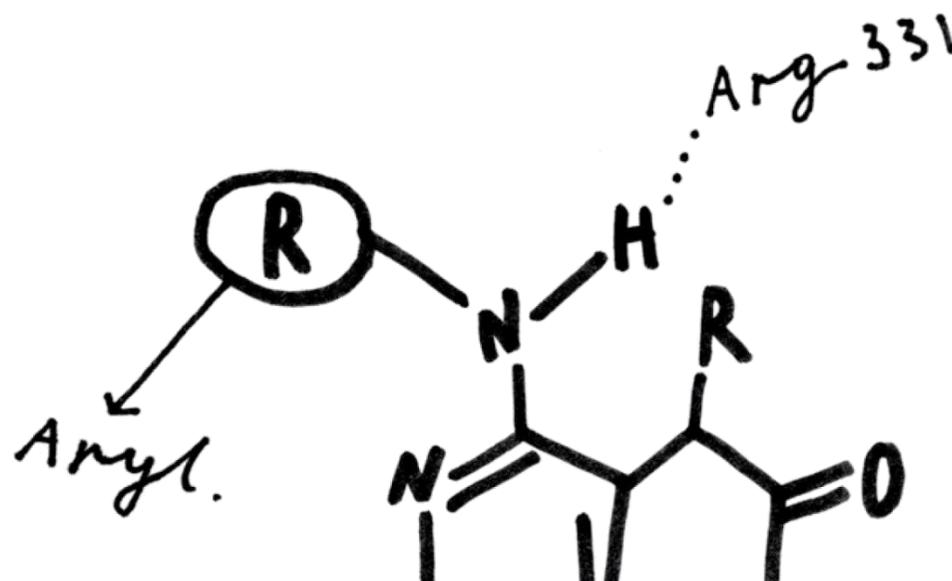
²Fedorov DG. et. al.; Acc. Chem. Res. 2014, 47, 2846–2856

³Li H., Fedorov DG. et. al.; J Comput Chem. 2010 Mar;31(4):778-90

Agenda

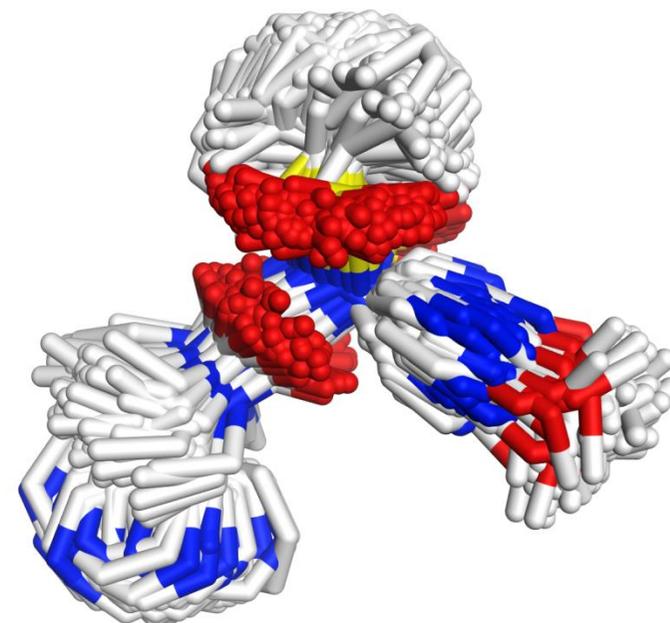
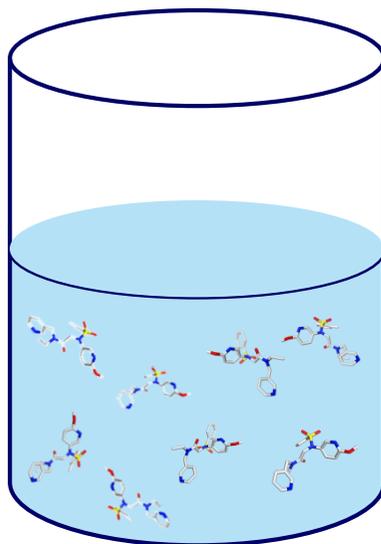
Fragment Molecular Orbitals (FMO) for GPCR exploration

HGMP-C4XD approach for GPCR drug discovery

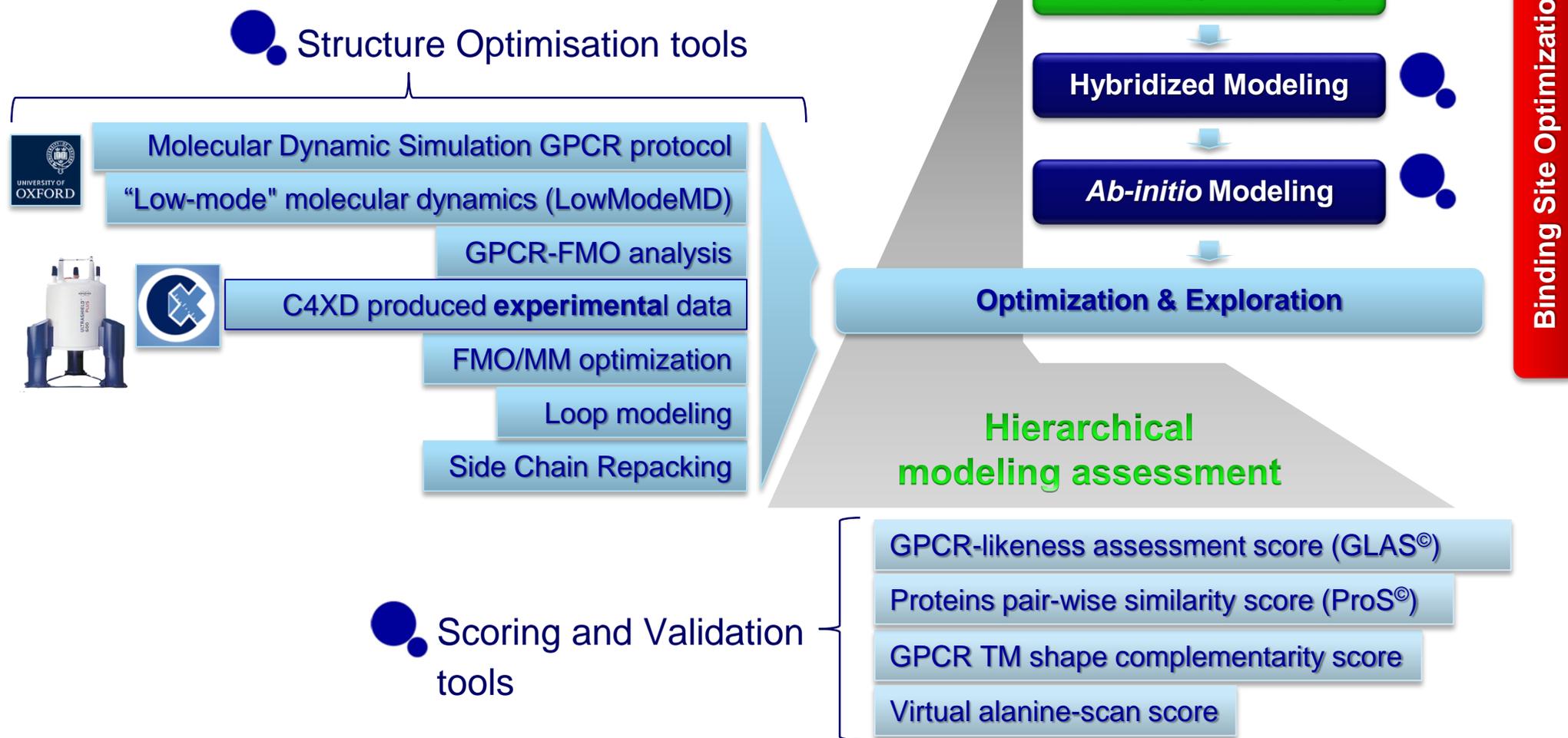


C4XD technology

- C4XD is an NMR-based technology to explore how molecules behave in physiological-relevant solution
- C4XD was developed by C4X Discovery Ltd¹
- C4X Discovery Ltd¹ experimentally demonstrated that small molecules exist in relatively few conformations in solution and that one of those conformations closely resembles the bioactive form – but which one?
- Can the C4XD technology identify the bioactive conformation of GPCR ligands?



Hierarchical GPCR modeling protocol (HGMP)¹⁻²



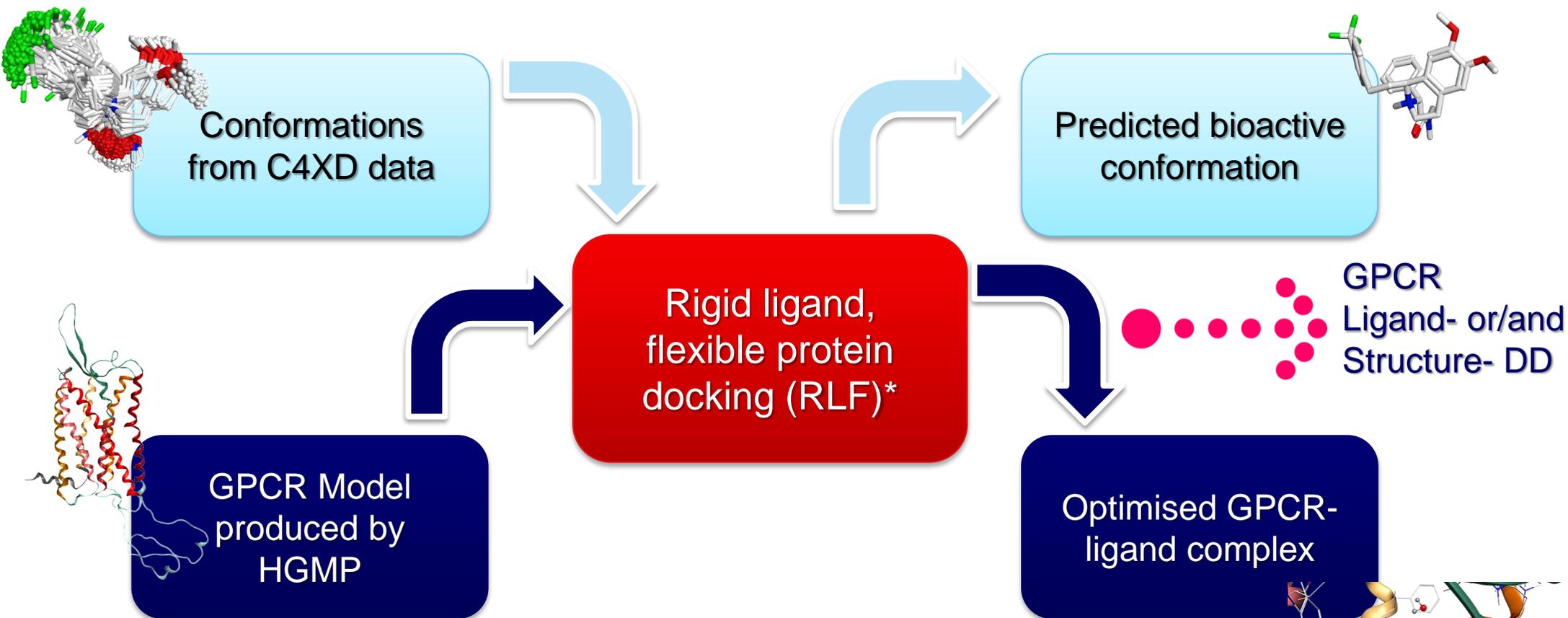
Prediction of the GPCR-ligand binding pose

Challenges

- The binding sites of GPCRs adopt their topology* for ligand binding
– this need to be addressed by GPCR modelling
- Most standard docking protocols keep the receptor (largely) rigid, and so do not address this flexibility
- Fully flexible (ligand and protein) docking can generate a lot of solutions that are difficult to separate
- By limiting the ligand to **experimentally determined** conformations and allowing receptor flexibility we are better able to address the ligand-induced GPCR flexibility

Prediction of GPCR-ligand complex

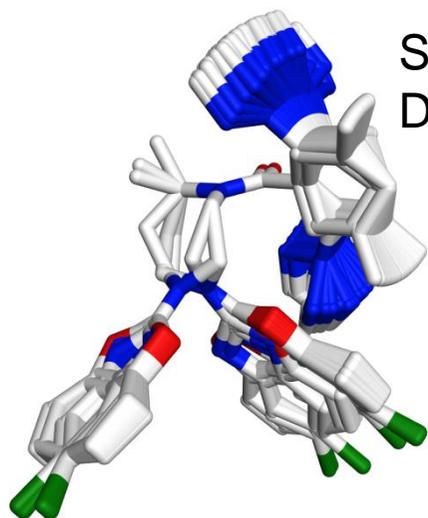
Workflow of HGMP-C4XD protocol



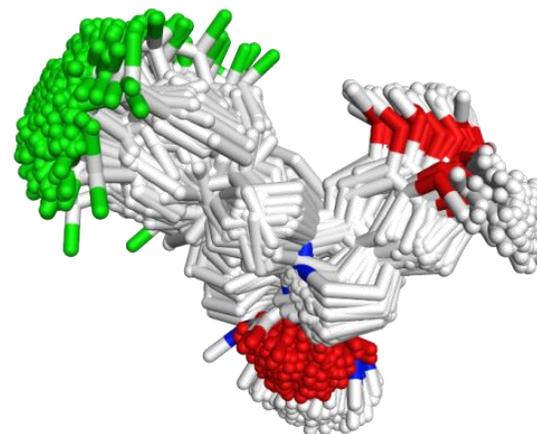
Rigid ligand, flexible protein docking protocol (RLF), implemented in the MOE v.2014.09 software package, treats the ligand as a rigid-body that allows it translational/rotational changes but keeps the dihedral angles fixed. The protein backbone and sidechains are flexible but the heavy atoms have distance restraints.

Designing OX-1 selective anatagonists

HGMP-C4XD

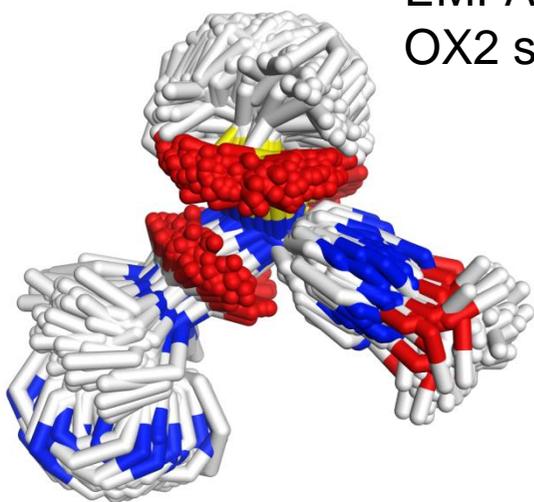


Suvorexant
Dual

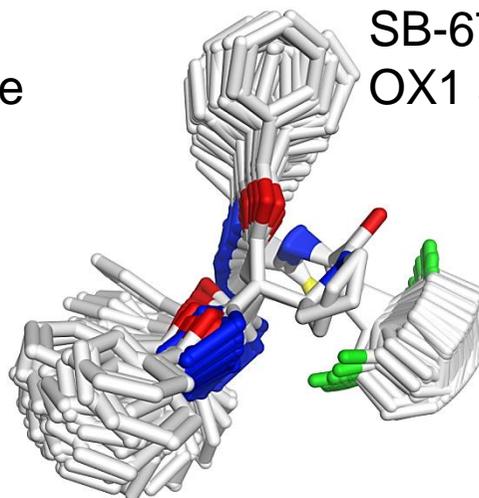


Almorexant
Dual

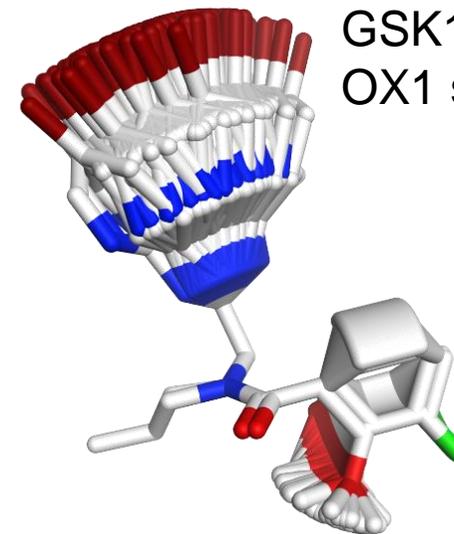
EMPA
OX2 selective



SB-674042
OX1 selective



GSK1059865
OX1 selective



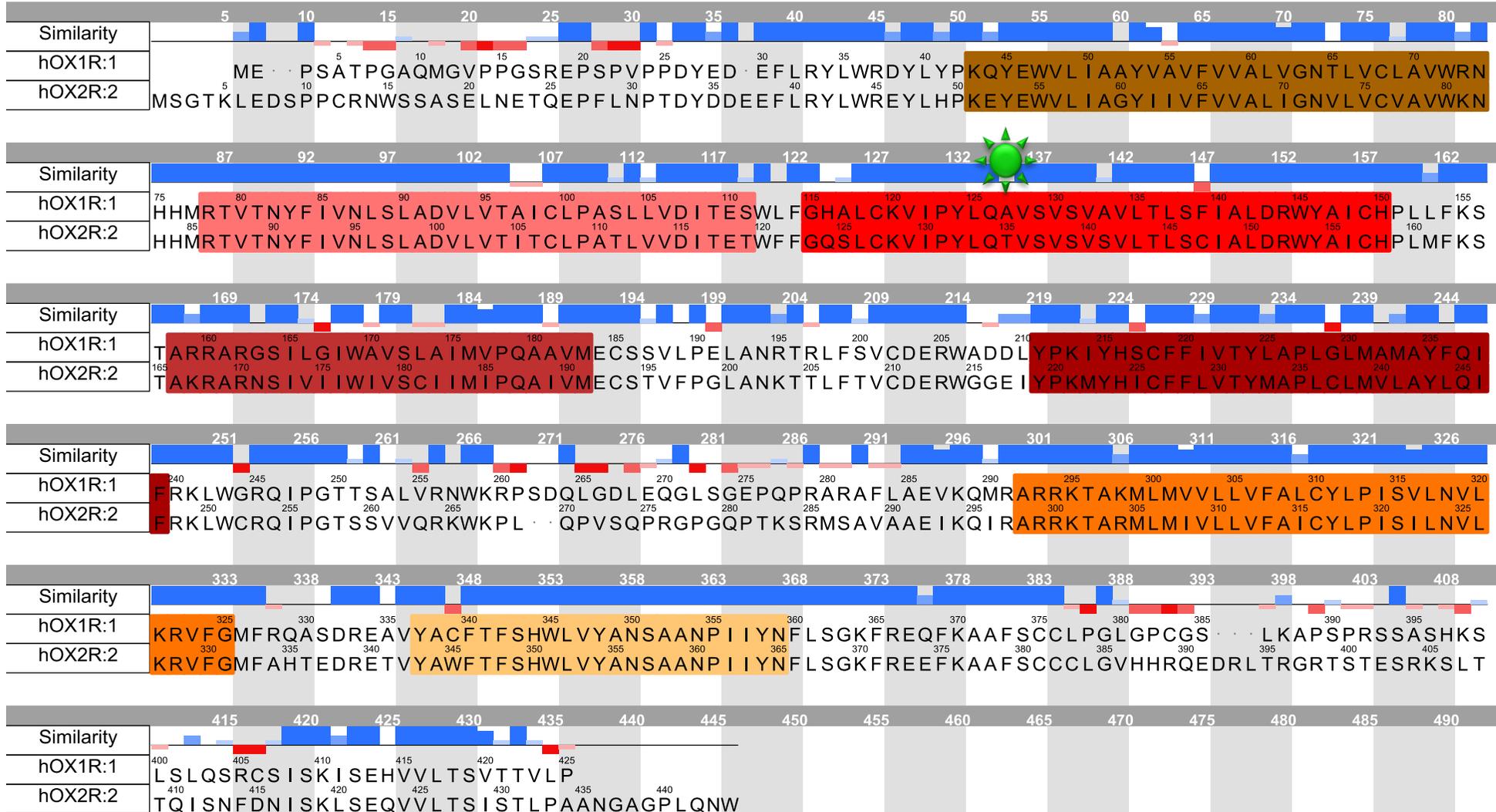
hOX1/hOX2 Sequence alignment

1:hOX1R

2:hOX2R

74.5
77.9

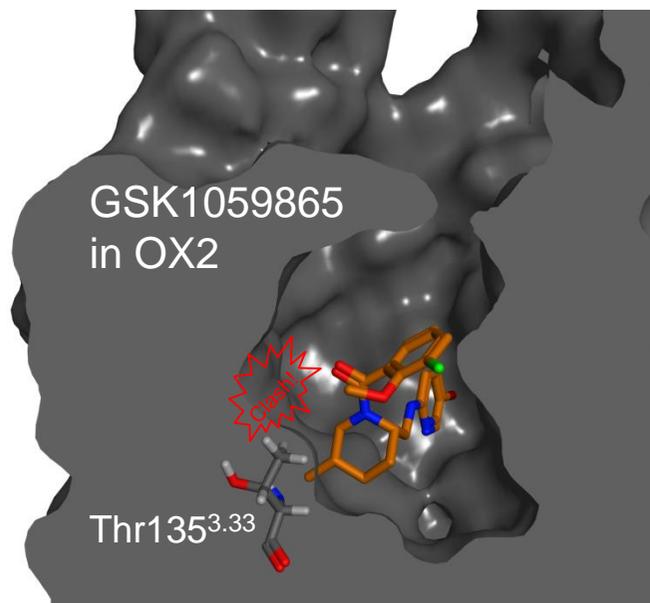
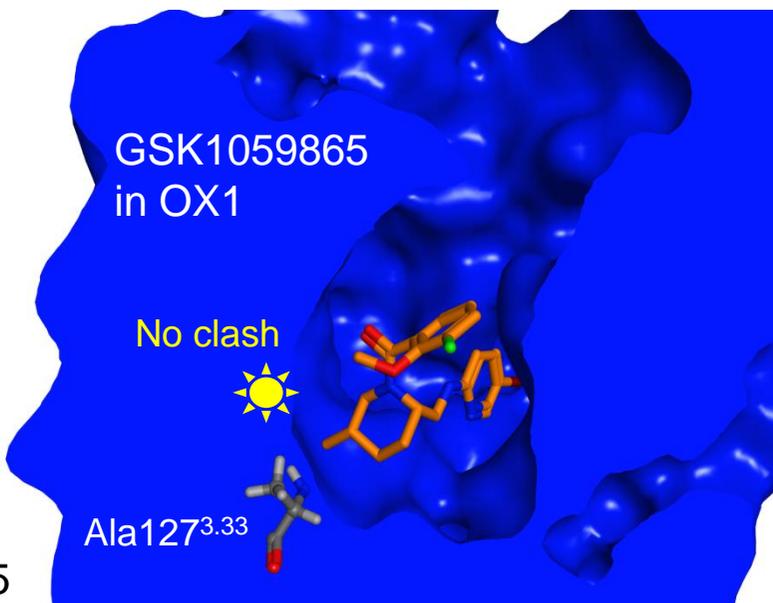
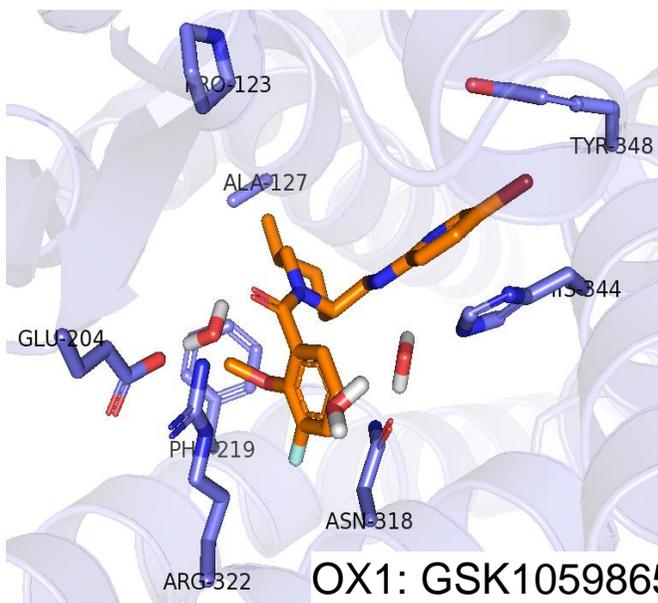
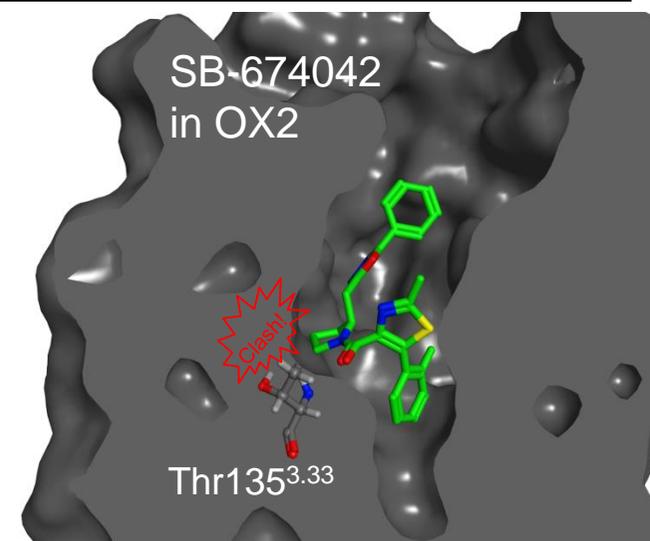
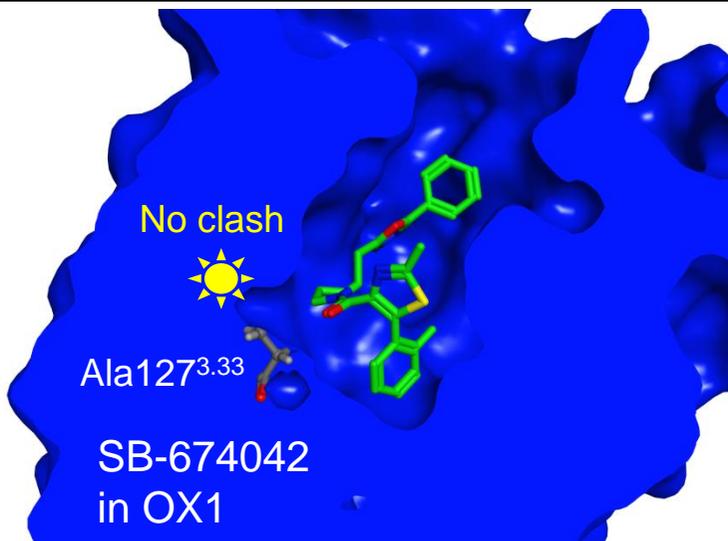
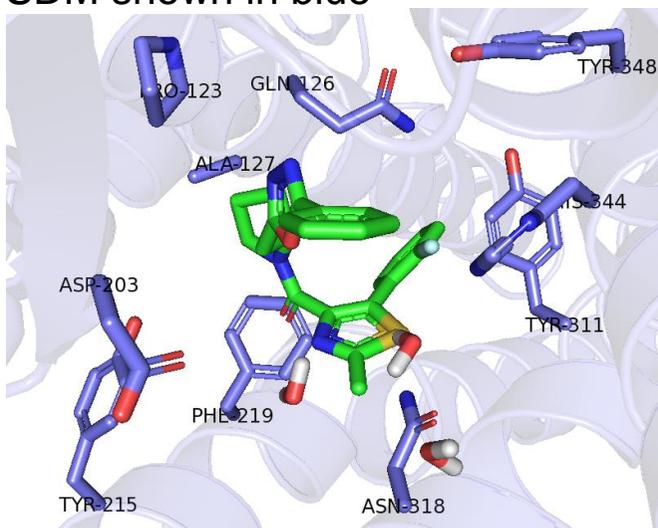
Similarity



OX1 Selective Antagonists

OX1: SB-674042
SDM shown in blue

RLF docking poses

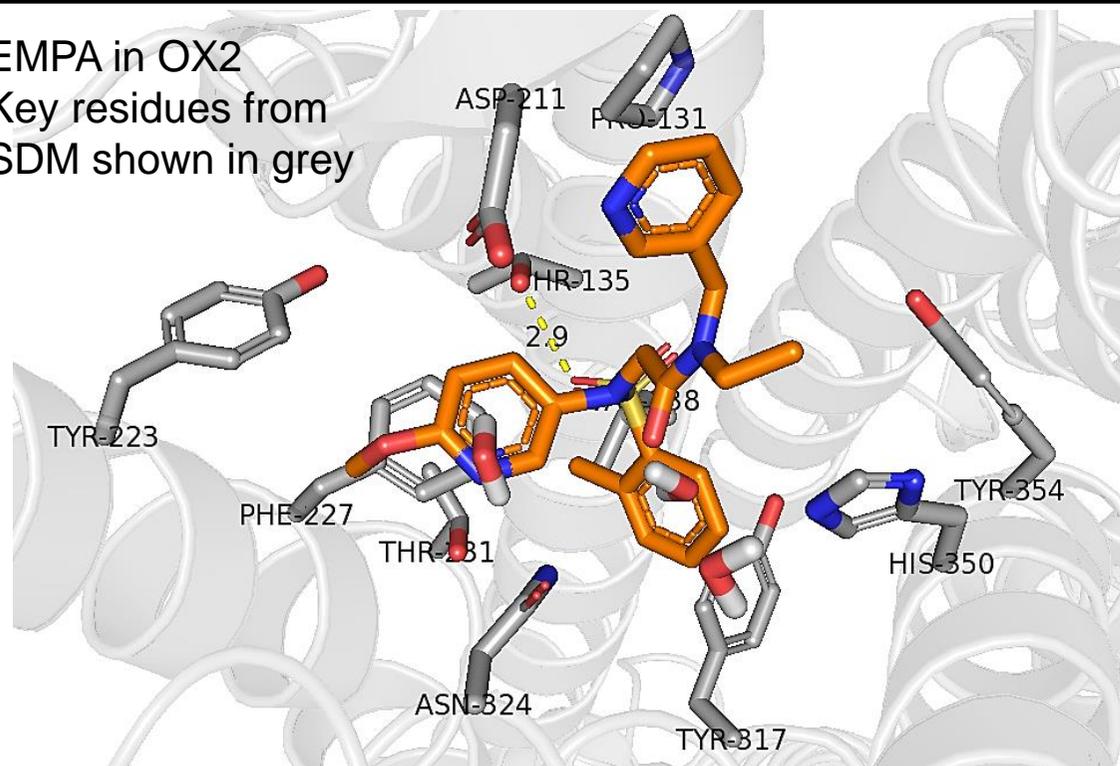


OX1: GSK1059865
SDM shown in blue

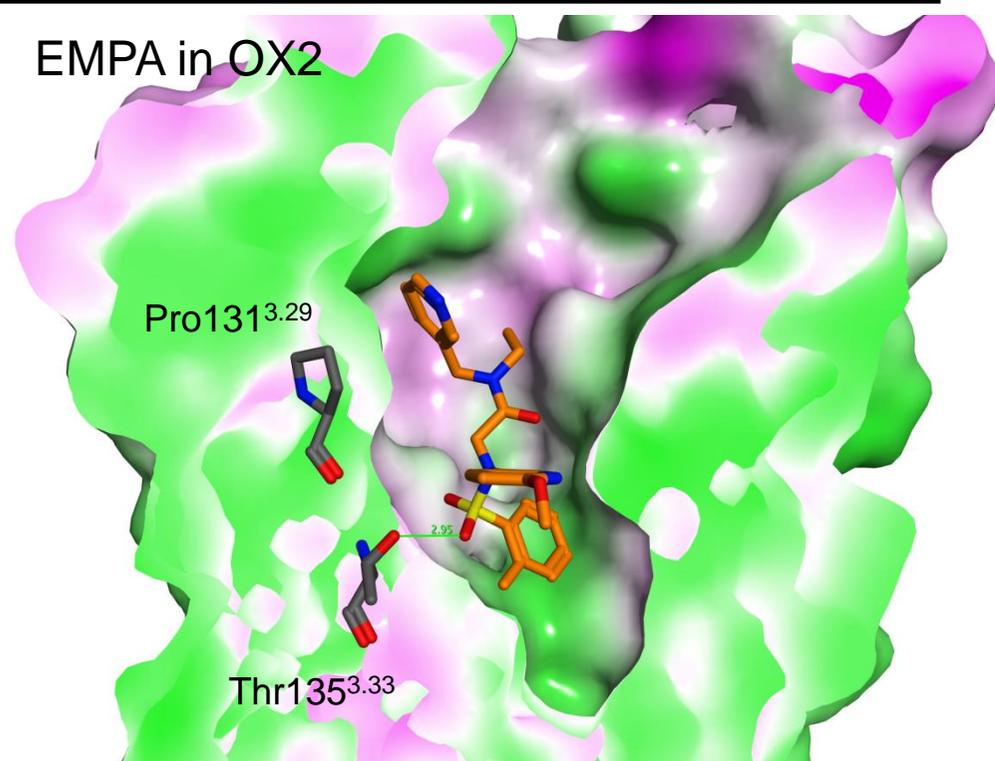
OX2 Selective Antagonist

EMPA

EMPA in OX2
Key residues from
SDM shown in grey



EMPA in OX2



- The docking pose of EMPA was validated by SDM data including the interaction with Thr135^{3.33}
- EMPA forms hydrogen bond with non-conserved Thr135^{3.33}, this interaction does not exist in OX1 – this can explain the OX2 selectivity of EMPA

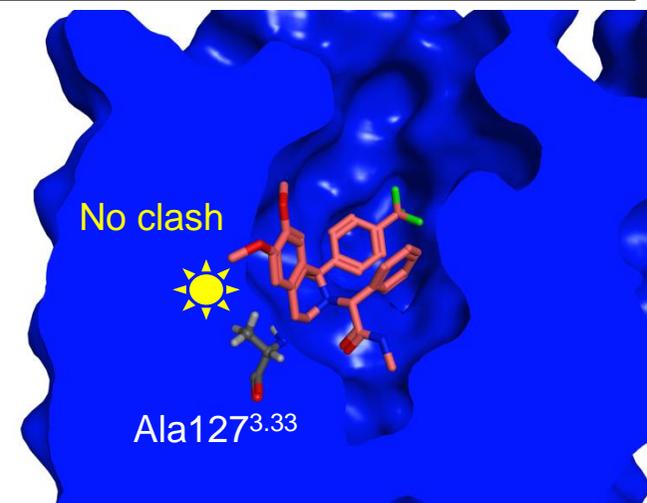
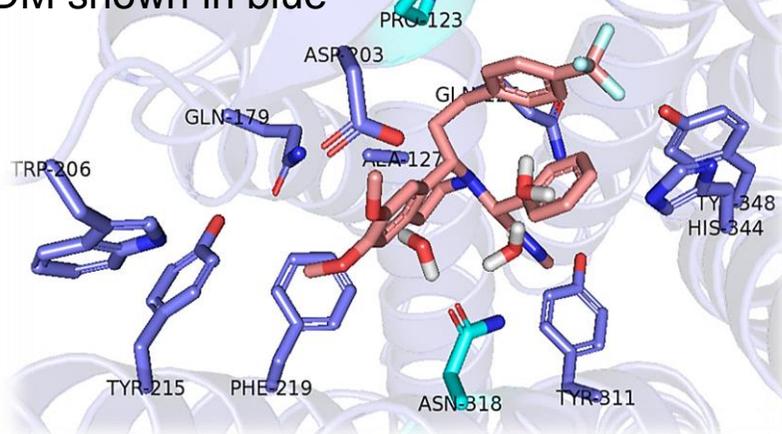


OX1/2 Dual Antagonists

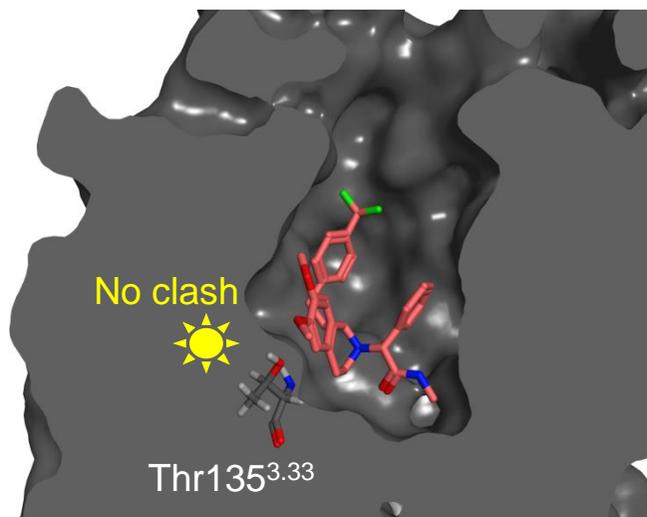
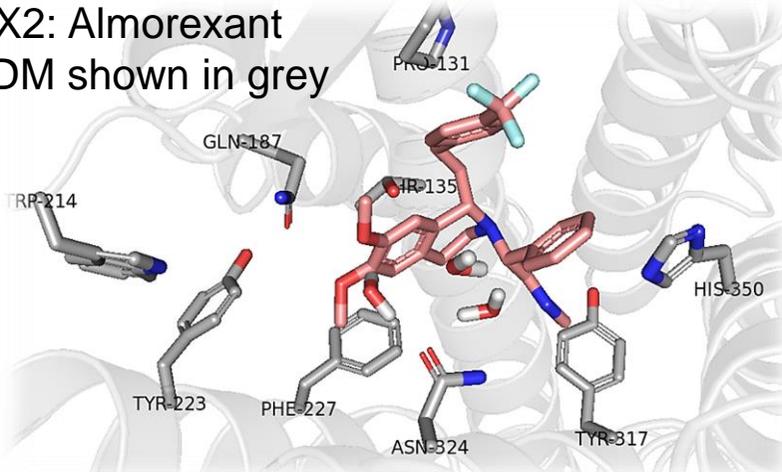
Almorexant

- The docking pose of Almorexant within OX1 and OX2 receptors
- The docking pose in both cases validated by site-directed mutagenesis data¹⁻²
- No clashes observed between Almorexant with OX2 or OX1 receptors that can explain its duality

OX1: Almorexant
SDM shown in blue



OX2: Almorexant
SDM shown in grey

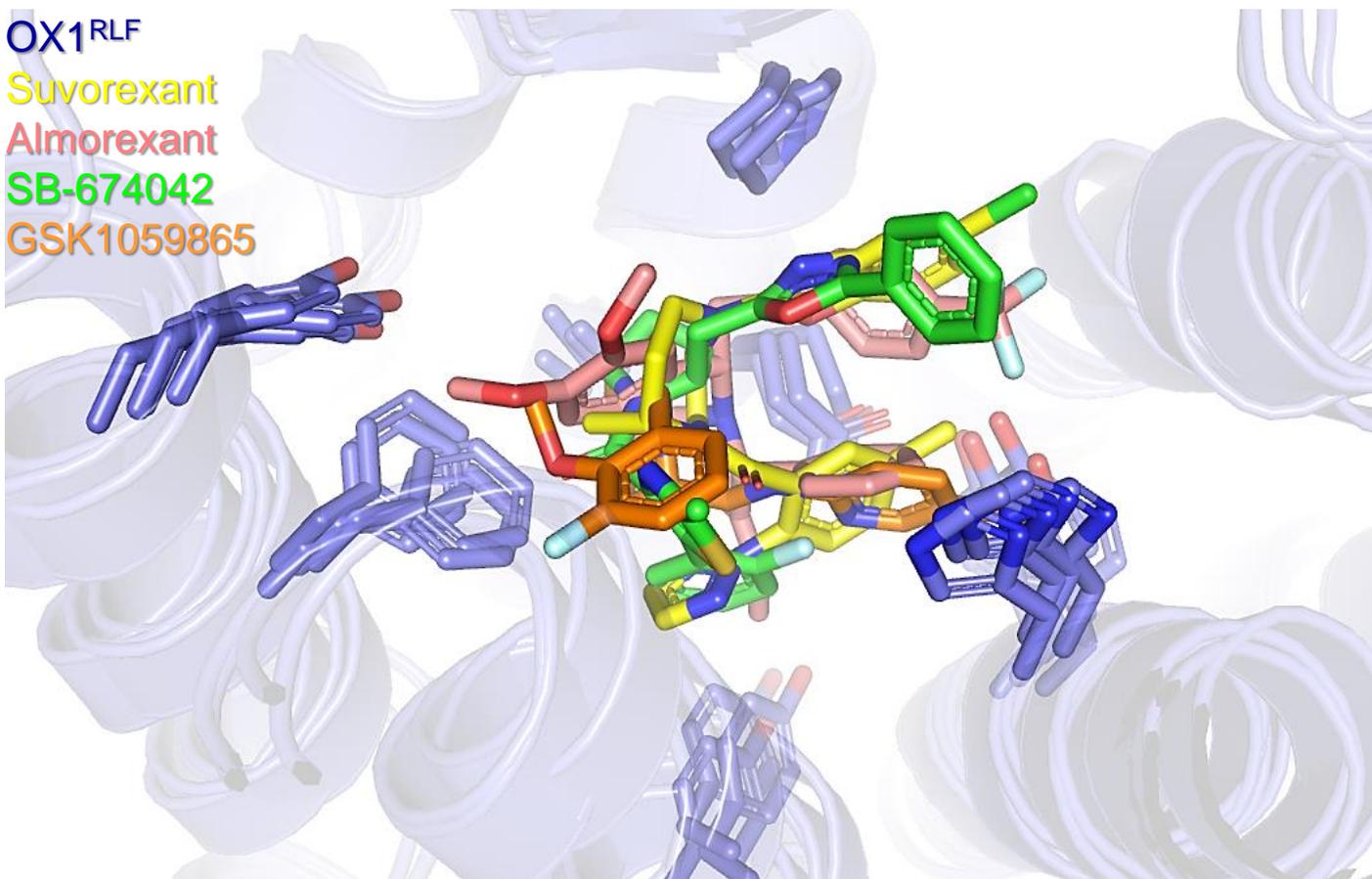


OX1 Antagonists

Addressing receptor flexibility

- RLF produced binding poses for key OX1 antagonists
- Different antagonists can access different areas of the OX1 pocket
 - The same phenomenon is observed in OX2
- Different ligand chemotypes induce different conformational changes in the OX1 binding site
- HGMP-C4XD (RLF) docking protocol addresses ligand-induced binding-site flexibility

OX1^{RLF}
Suvorexant
Almorexant
SB-674042
GSK1059865



HGMP-C4XD

Preliminary conclusions and working plan

- The combination of C4XD data and HGMP was very efficient in predicting the binding poses of 5 dual and selective OX1/2 antagonists
- The HGMP-C4XD protocol can isolate the bioactive conformation from the ensemble of ligand conformations in solvent
- The bioactive conformation was found amongst the most populated conformers in the solvent
- The structural insights produced by this approach are very useful in further SBDD of OX1 selective antagonists
- Further validation of the method is required/ongoing

Acknowledgements

University of Oxford

Matteo Aldeghi

Dr. Philp C. Biggin

AIST Tsukuba Japan

Dr. Dmitri Fedorov

Evotec

Dr. Ewa Chudyk

Dr. Laura Gleave

Dr. Tim James

Dr. Tara Fryatt

Dr. Jonathan Bentley

Dr. Mike Bodkin

C4X Discovery

Dr. Thorsten Nowak

Dr. Barrie Martin

Your contact:

Alexander Heifetz
Principal Scientist

+44.(0) 1235 83 8925
alexander.heifetz@evotec.com

