



smarter chemistry | smarter decisions™

## Bioisosteres in Accessible Chemistry Space

Tim Cheeseright

# About Cresset

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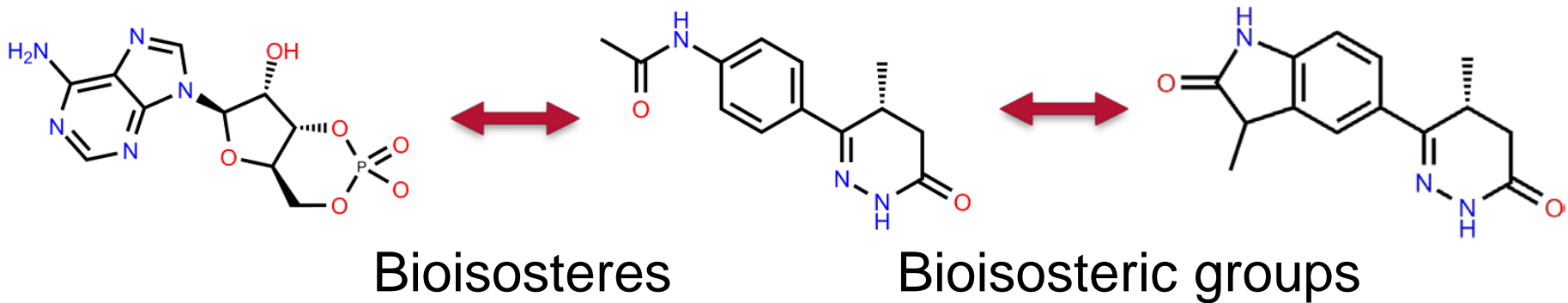
- > Founded in 2002 by Dr. Andy Vinter – initial Wellcome Trust funding
- > Located in Cambridge, UK
- > Primary market pharmaceutical and biotech R&D
  - > Other markets include flavour and fragrance, agrochemical and chemicals
- > **Software:**
  - > 14 of the top 20 pharmaceutical companies use Cresset technology in their research programmes
  - > First software sale 2005
- > **Consultancy Services:**
  - > 160 collaborative projects delivered to global clients

# What are Bioisosteres?

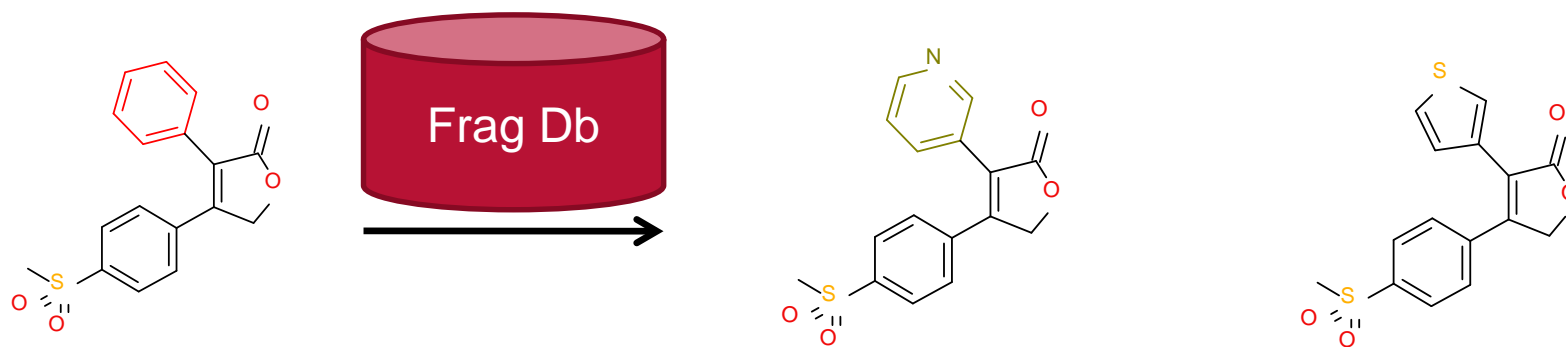
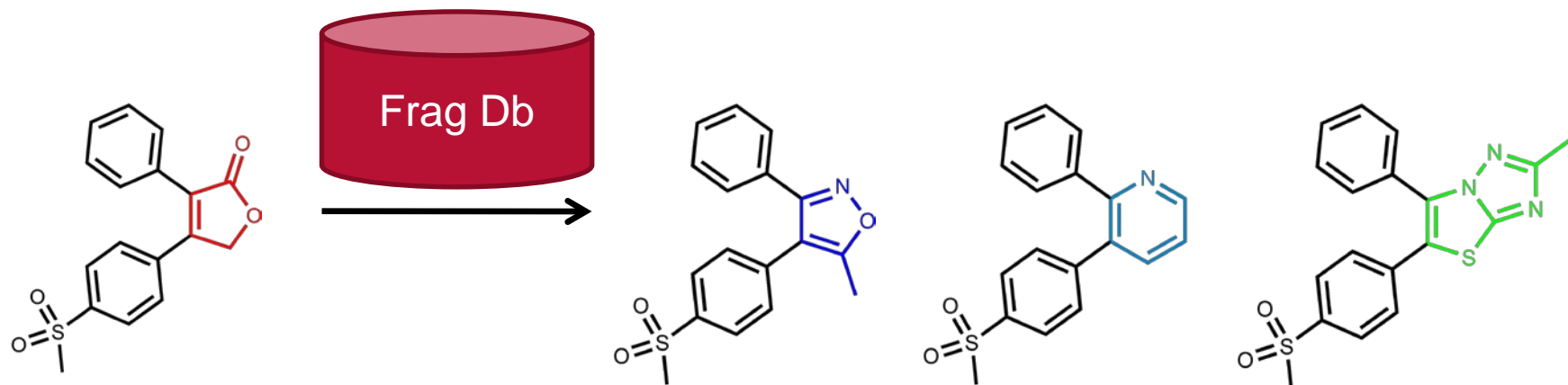
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> Broadly:

Any two structures that show the same biological effect



# Searching for Bioisosteric Groups



# Scoring Replacements

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- > **Structure Based**

- > Docking

- > **Ligand Based**

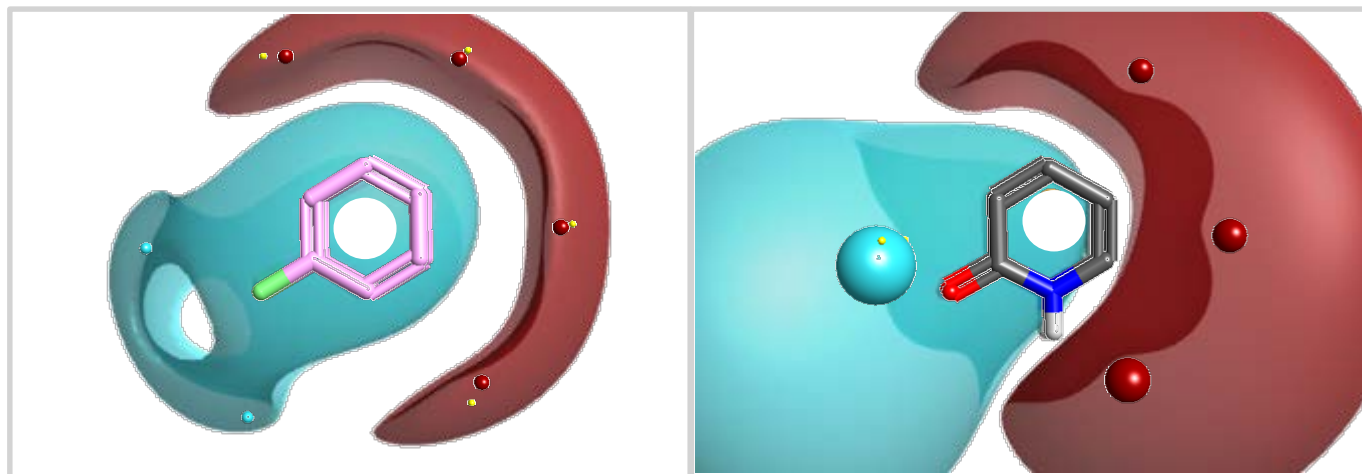
- > 2D similarity

- > Pharmacophores

- > Electrostatic and Shape Similarity

# Alignment, Scoring and Comparisons

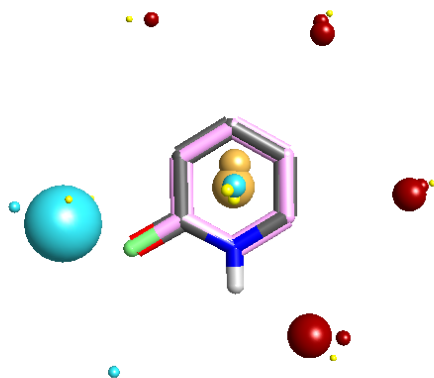
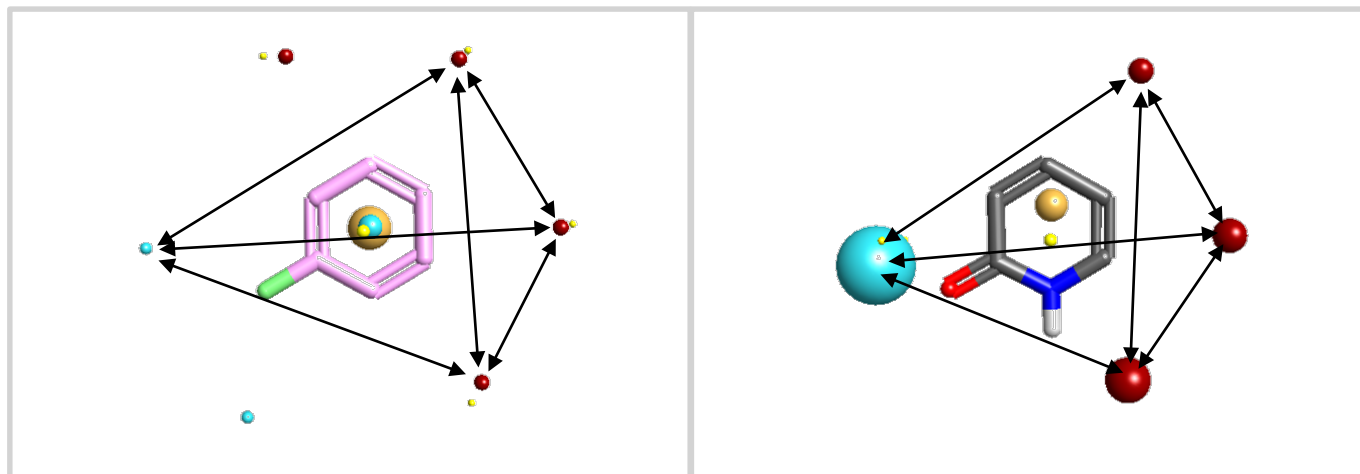
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# Alignment, Scoring and Comparisons

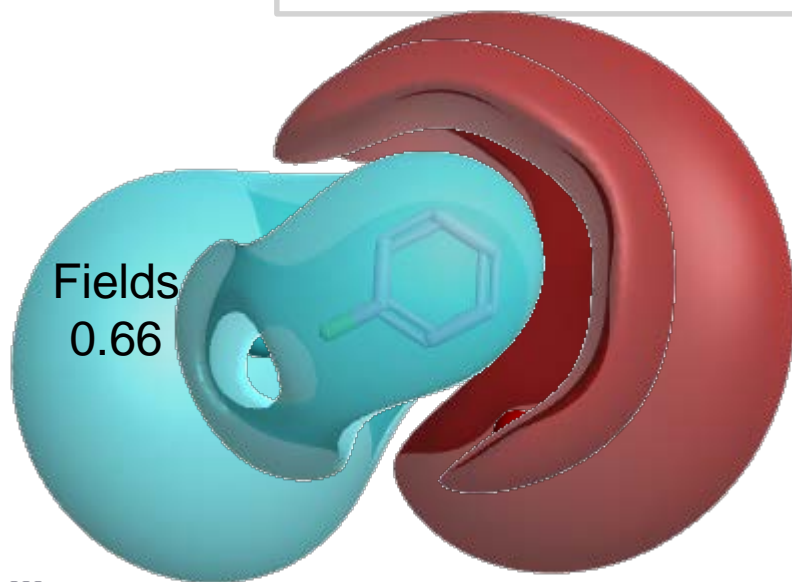
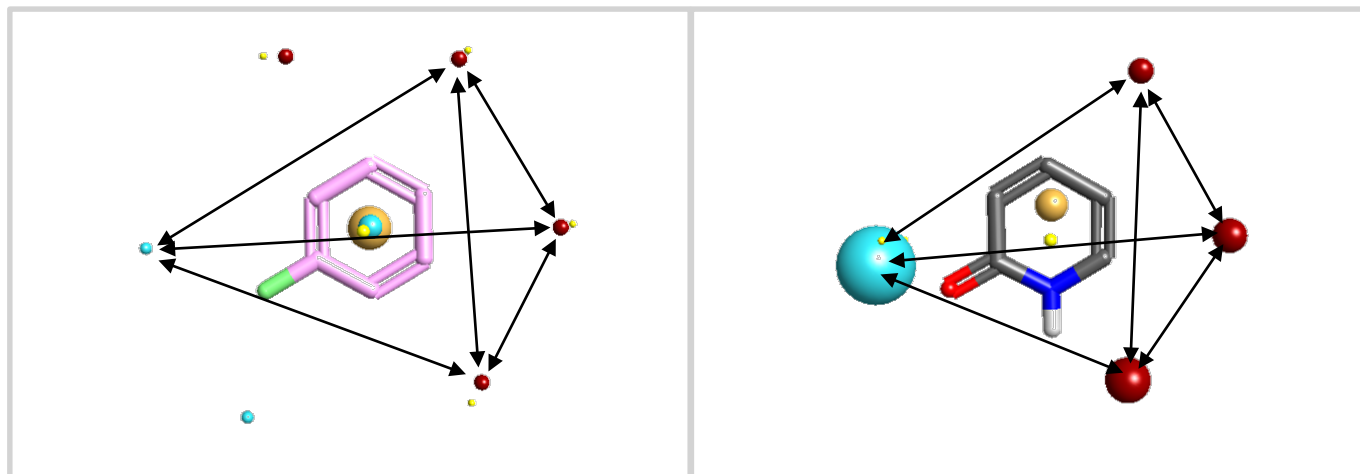
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Clique based alignment



# Alignment, Scoring and Comparisons

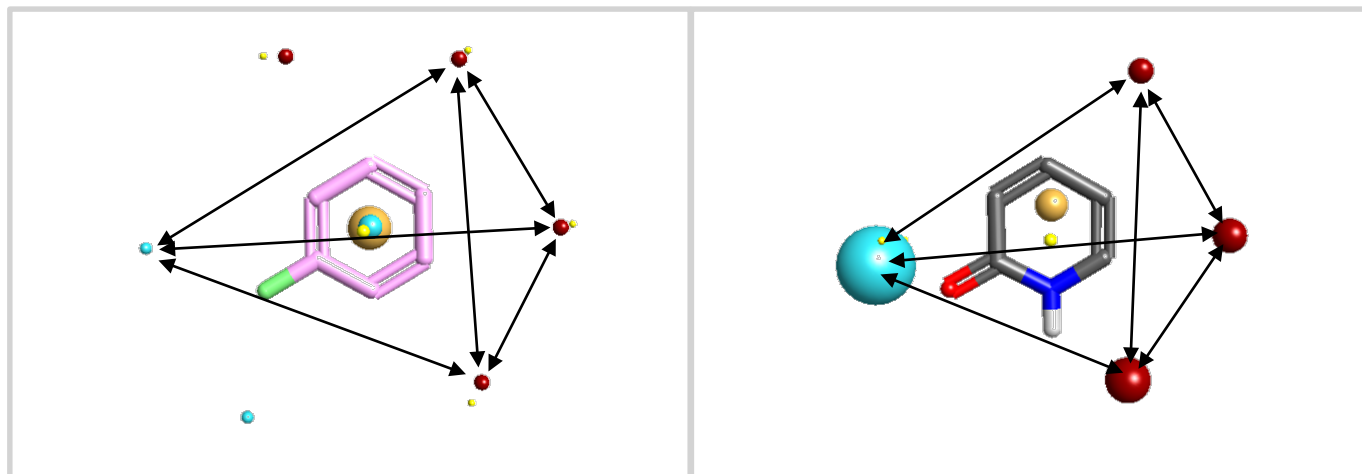
Clique based alignment



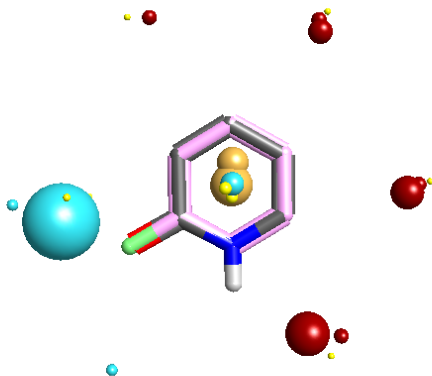


# Alignment, Scoring and Comparisons

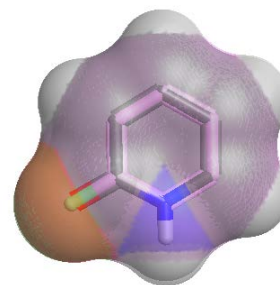
Clique based alignment



Fields  
0.66



Cheeseright et al,  
*J. Chem Inf. Mod.*, 2006, 665

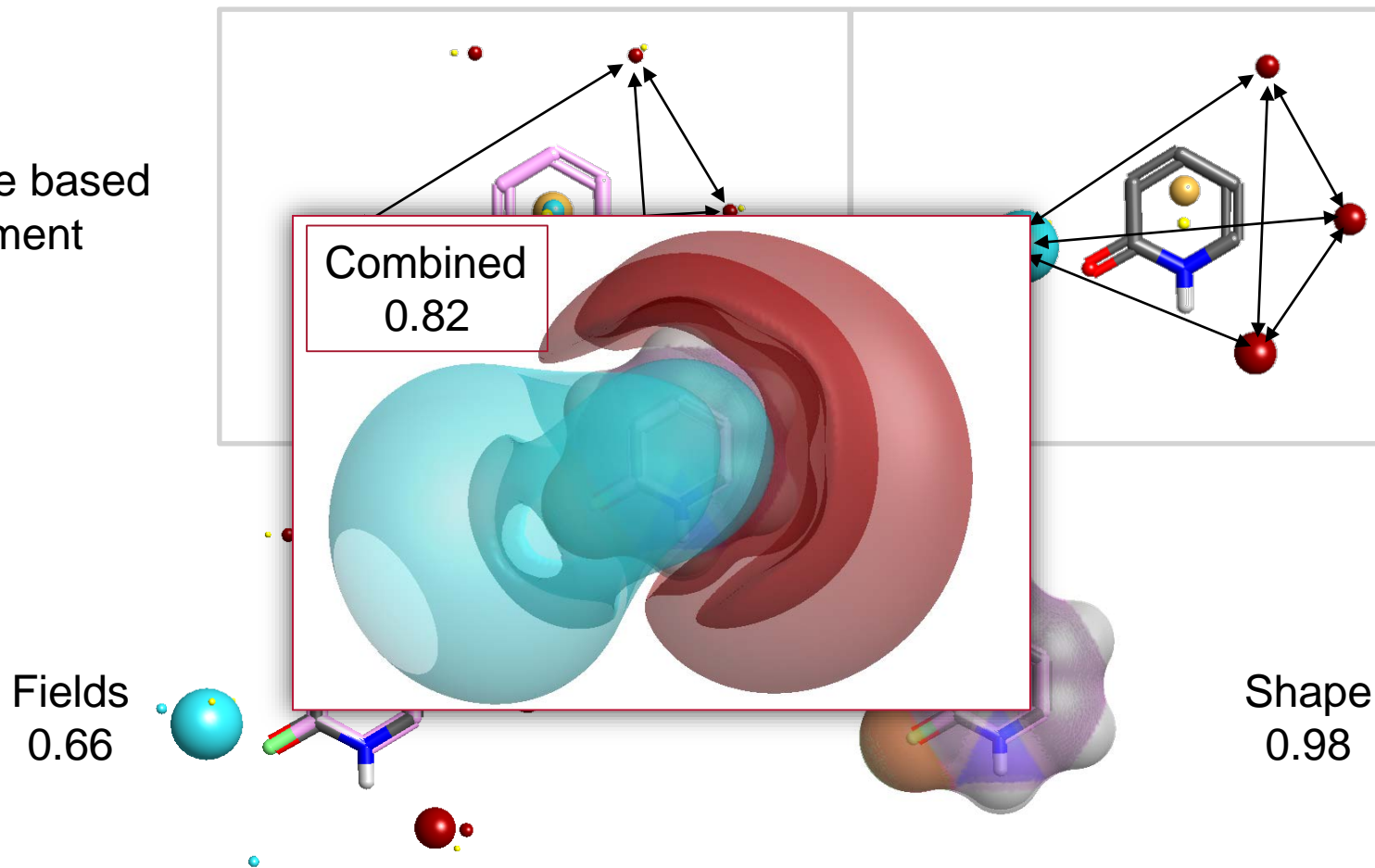


Shape  
0.98

Grant, Gallardo, Pickup,  
*J. Comp. Chem.*, 1996, 1653

# Alignment, Scoring and Comparisons

Clique based alignment

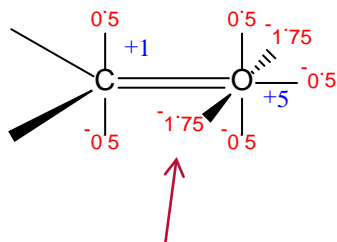


Cheeseright et al,  
*J. Chem Inf. Mod.*, 2006, 665

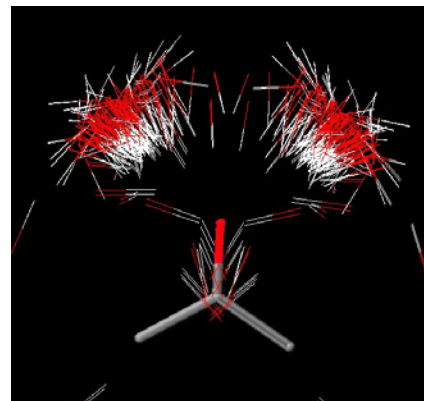
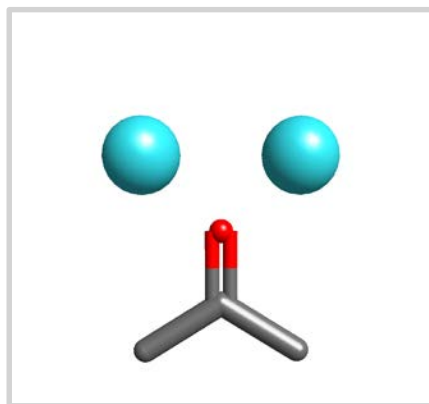
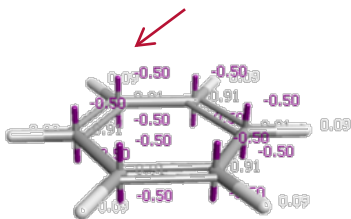
Grant, Gallardo, Pickup,  
*J. Comp. Chem.*, 1996, 1653

# Detailed Electrostatics from XED

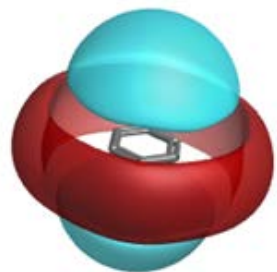
> eXtended Electron Distribution gives detailed electrostatic interaction patterns



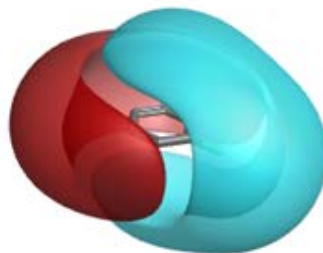
XED adds p-orbitals to get detailed representation of atoms



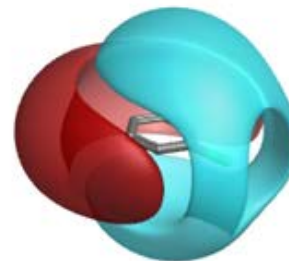
Interaction of Acetone and Any-OH from small molecule crystal structures



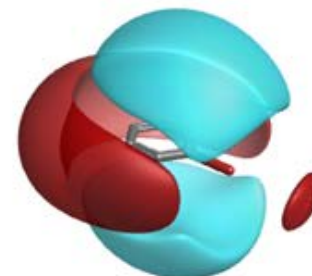
benzene



fluorene



chlorobenzene



bromobenzene

# Scoring with Shape and Electrostatics

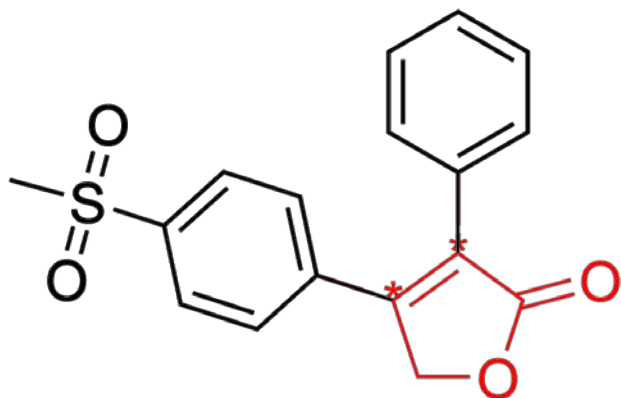
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- > Conformation dependant
- > Electrostatics are a property of the whole molecule
- > Shape is a property of the whole molecule

# Our Approach

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- 1 Select a region to replace and remove these atoms

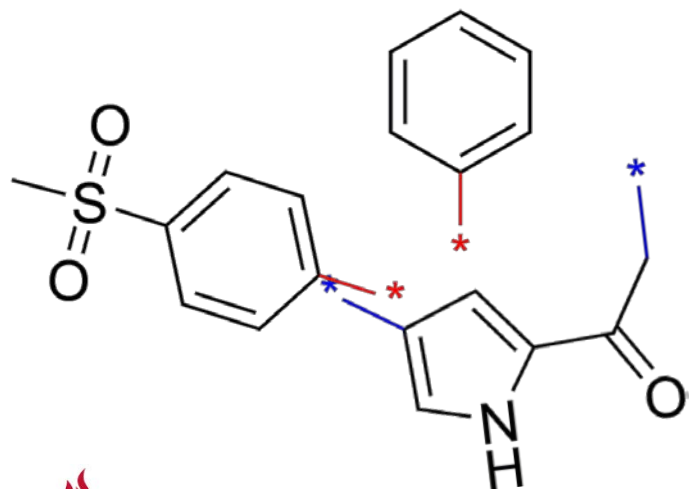


# Our Approach

---

- 1 Select a region to replace and remove these atoms
- 2 Search database for matching fragments  
(geometric search only)  
(search runs on fragment conformations)

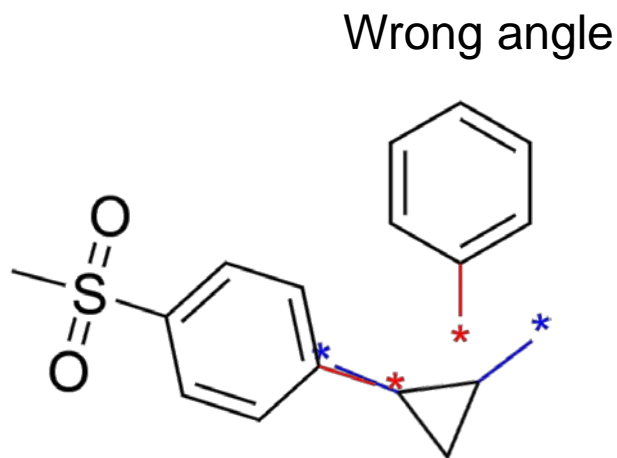
Wrong distance



# Our Approach

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- 1 Select a region to replace and remove these atoms
- 2 Search database for matching fragments  
(geometric search only)  
(search runs on fragment conformations)

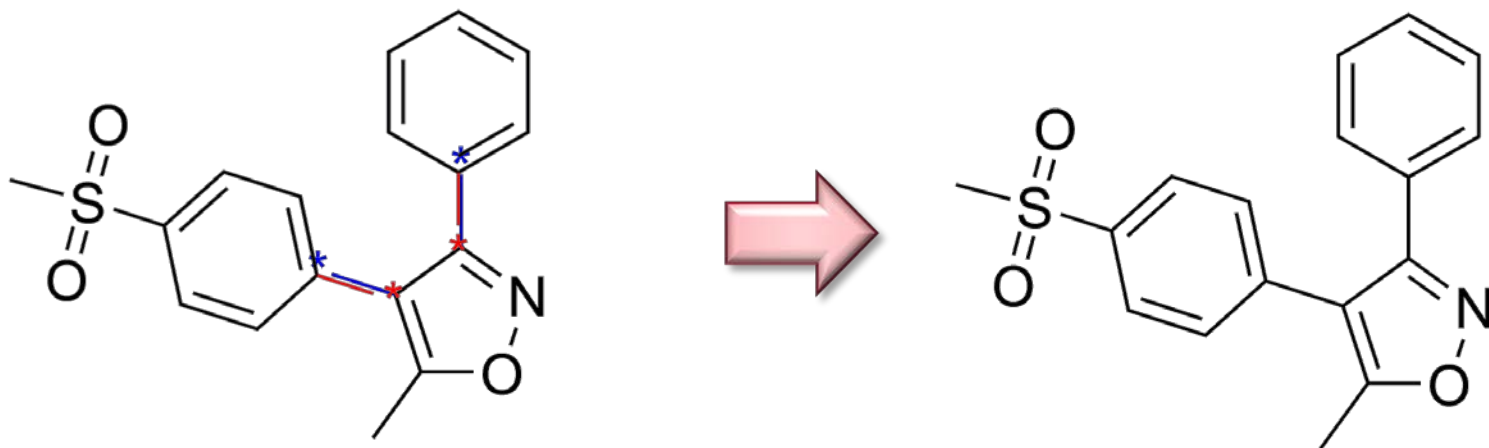


# Our Approach

---

- 1 Select a region to replace and remove these atoms
- 2 Search database for matching fragments  
(geometric search only)  
(search runs on fragment conformations)
- 3 Form Products  
(minimise and add Field Points)

Good match

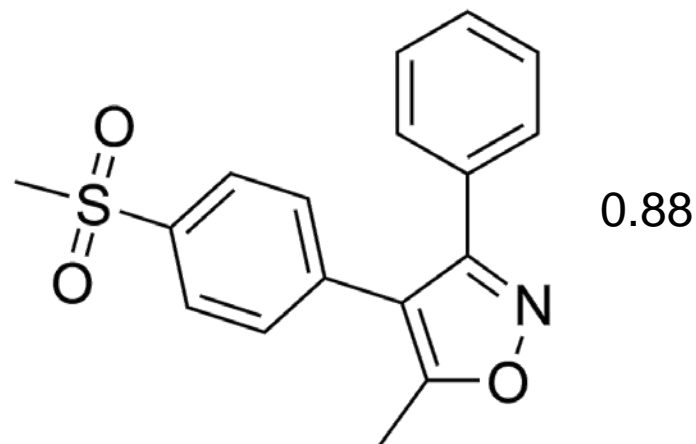
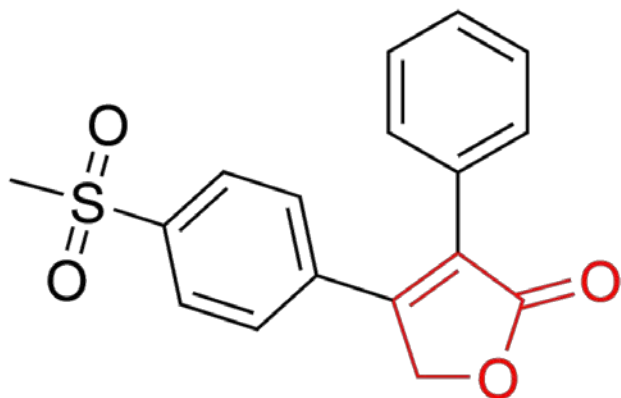




# Our Approach

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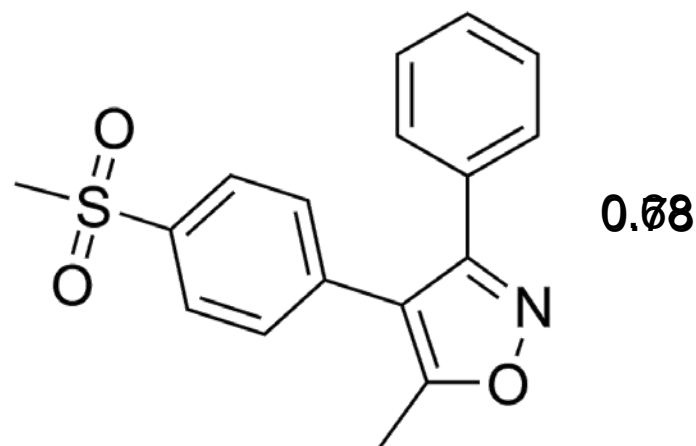
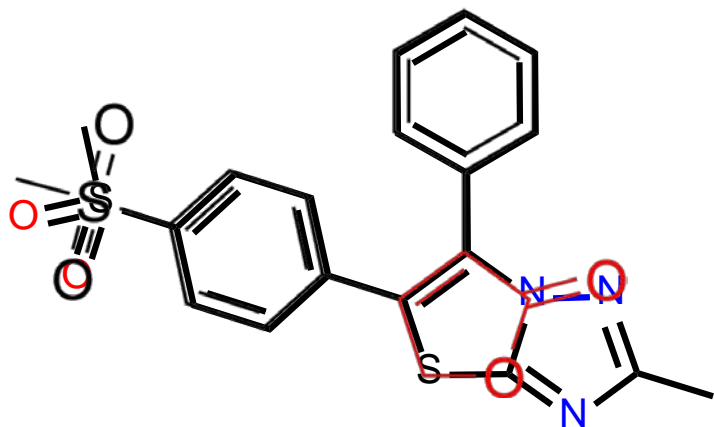
- 1 Select a region to replace and remove these atoms
- 2 Search database for matching fragments  
(geometric search only)  
(search runs on fragment conformations)
- 3 Form Products  
(minimise and add Field Points)
- 4 Score



# Cross scoring modification

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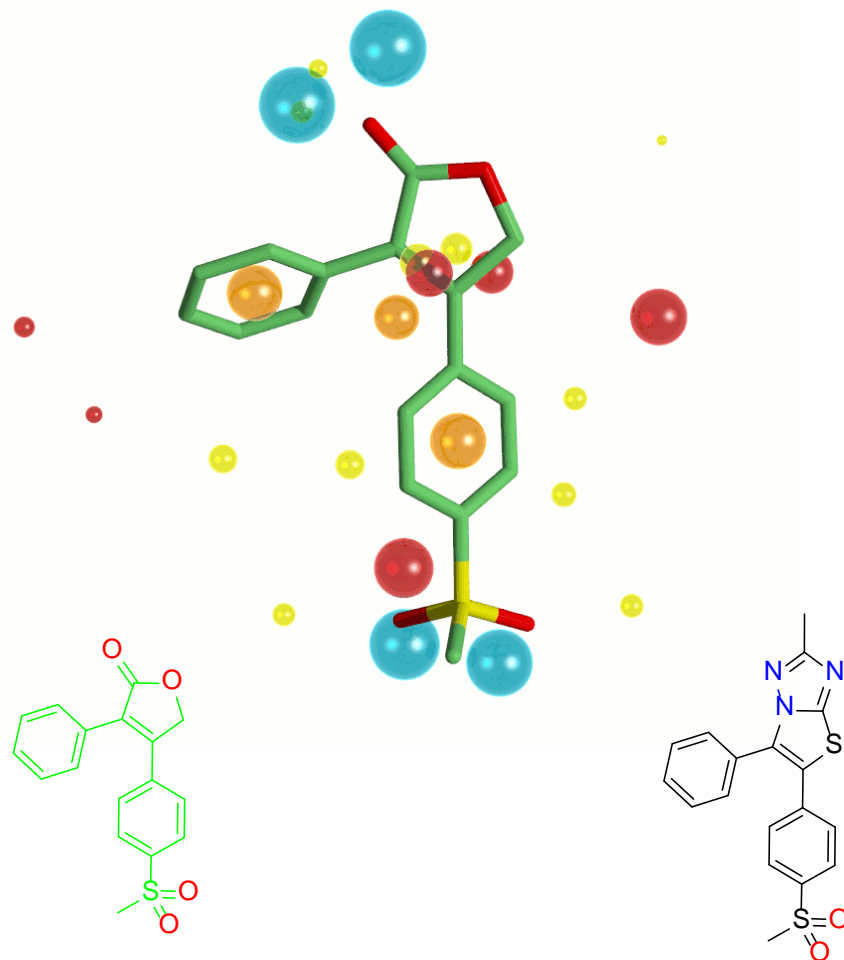
- 1 Select a region to replace and remove these atoms
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- 3 Form Products  
(minimise and add Field Points)
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# Whole-Molecule Scoring Advantages

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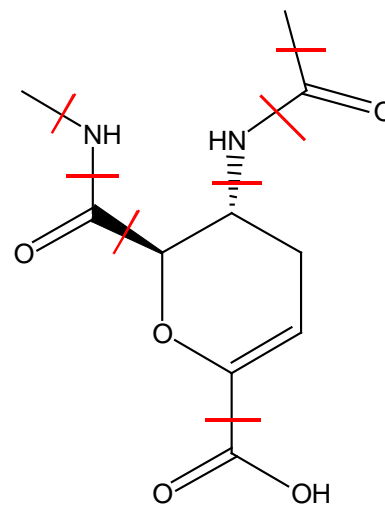
- > Produces more diverse, non-obvious bioisosteres
  - > Avoids fragment scoring limitations
  - > Allows for electronic influence of replacing a moiety on the rest of the molecule and vice versa
  - > Allows for neighbouring group effects



# Where do you get fragments from?

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- > Fragment existing molecules
- > Fragment at
  - > Heteroatoms
  - > C=O and C=S
  - > Bonds to rings
  - > Not NO<sub>2</sub>, COOH, etc
- > Generate all sets of connected pieces
  - > Subject to MW, NH, and rotatable bond limits
- > Sort fragments on frequency
  - > Zinc derived → 'VeryCommon', 'Common', 'LessCommon' etc
  - > ChEMBL derived → 'ChEMBL\_common' etc



# Core vs Terminal Replacements

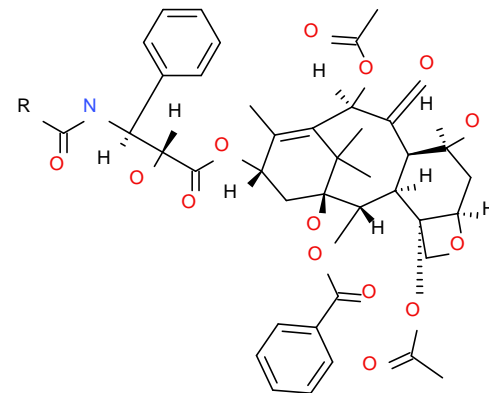
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- > Core replacement highly utilized in Cresset customers
  - > Algorithm supports both core and terminal replacements
    - > Terminal fragments require rotation after joining to find optimal alignment
  - > Customer request to link replacements heavily to available chemistry space more tightly
- Find terminal replacements from list of available reagents

# Processing Reagents into Databases

> Intended to aid synthetically accessible choices for a series

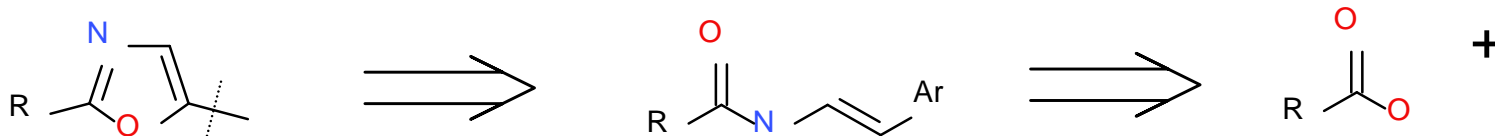
> Not assessing global “Synthesizability”



> Want to capture the origins of an R group

> Not trying to encode all chemical transformations

> Transformation can represent multiple chemical steps



# Atpat – Atom/pattern matching

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- > Cresset's Cheminformatics platform
- > Derived from (and similar to) Merck's internal PCP program
- > "AWK for Molecules"
- > Used extensively within Cresset software to process and modify molecules
  - > Fragmentation
  - > Re-charging
  - > Tautomerization
  - > Calculations of TPSA, logP etc

# More on atpat

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- > Pattern/action language

- > Syntax:

PATTERN COMMAND

- > Pattern can be smarts like atom and/or bond pattern or property that is held by atoms/bonds

- > Commands or Actions can operate on

  - > Atomic - change atom, charge, delete, add

  - > Bond - change type etc

  - > Residue

  - > Molecule based



# Advantage over SMARTS

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## > Flexibility

> Assign properties to atoms and match these properties in patterns

> Multi-line processing enables simple understanding of complex patterns

> E.g. Definition of a protonatable amine

> Recursive SMARTS (Wikipedia):

```
[ $( [NH2] [CX4] ) , $( [NH] ( [CX4] ) [CX4] ) , $( [NX3] ( [CX4] ) ( [CX4] ) [CX4] ) ]
```

> ATPAT:

```
N&sp3~sp2 | sp | N      > Ntri _ ;
```

```
N&sp3!Ntri              > basicN ;
```

# Disadvantages of ATPAT

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- > “What’s ATPAT?”
- > Learning curve
- > Odd syntax

# Reagent Processing using Atpat

Fragmentation method: Fragment settings | Fragment size | Conformation hunt

Fragmentation mode: Reagent importer

Reagent type: Acids/acid chlorides, delete the -COOH

Acids where we keep only the group attached to the acid carbonyl.  
e.g. R-COOH -> R-\*

$$\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH} \longrightarrow \text{R}-^*$$

```
{
TITLE Acids/acid chlorides, delete the -COOH
DESCRIPTION Acids where we keep only the group attached to
the acid carbonyl.DESCRPTION e.g.  R-COOH -> R-*
}
!H~C&c3&sp2(~O&c1)~O|F|Cl|Br|I&c1
$n _ 52      _      _
> _ _      del  " ;
del  > # ;
```

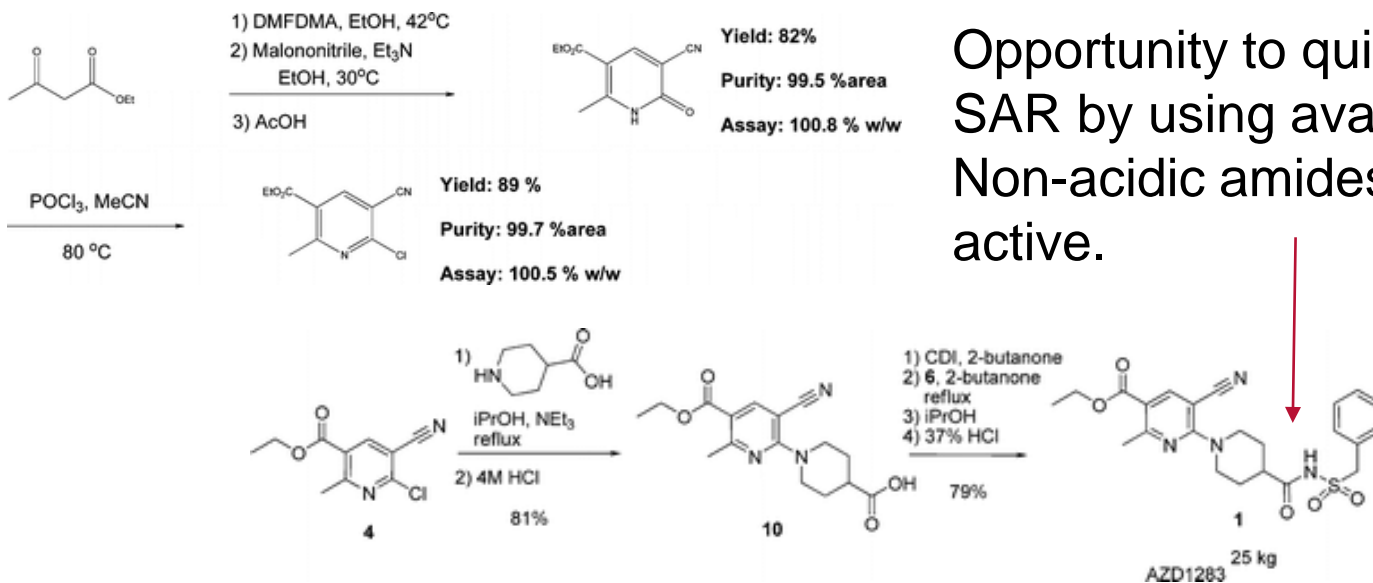
# Current implementation

- > 17 rules for conversion of R groups to fragments
- > Only single attachment points
  - > Leaf groups not cores
- > <http://cresset-group.com/sparkreagentdb/>

Transformation	Title	Description
$\text{R}-\text{C}(=\text{O})\text{OH} \longrightarrow \text{R}-\text{C}^*$	Acids/acid chlorides, delete the -COOH	Acids where we keep only the group attached to the acid carbonyl. e.g. R-COOH -> R-*
$\text{R}-\text{C}(=\text{O})\text{OH} \longrightarrow \text{R}-\text{C}(=\text{O})-\text{C}^*$	Acids/acid chlorides, keep the carbonyl	Acids where we attach through the carbonyl group (eg acylations) e.g. R-COOH -> R-C(=O)-*
$\text{R}-\text{OH} \longrightarrow \text{R}-\text{C}^*$	Aliphatic alcohols, delete the O	Alcohols used as alkylating agents where the O is deleted on addition e.g. R-OH -> R-*
$\text{R}-\text{OH} \longrightarrow \text{R}-\text{O}-\text{C}^*$	Alcohols, keep the O	Alcohols where the attachment is through the oxygen e.g. R-OH -> R-O-*
$\text{R}-\text{Cl/Br/I} \longrightarrow \text{R}-\text{C}^*$	Aliphatic halide	Primary/secondary/tertiary aliphatic halides (Cl,Br,I) e.g. R(1-3)C-Cl -> R(1-3)C-*
$\text{R}-\text{C}\equiv\text{C} \longrightarrow \text{R}-\text{C}^*$	Alkynes, delete the -C#C	Alkynes, keep only the attached group e.g. R-C#C -> R-*
$\text{R}-\text{NH}_2 \longrightarrow \text{R}-\text{C}^*$	Amines, delete the N	Primary amines as an alkylating agent where the N is deleted on addition e.g. R-NH2 -> R-*
$\text{R}-\text{NH}_2 \longrightarrow \text{R}-\text{N}(\text{H})-\text{C}^*$	Amines, keep the N	Primary and secondary amines where the N is the attachment point such as in reductive aminations e.o.

# Case Study – Application to P2Y<sub>12</sub>

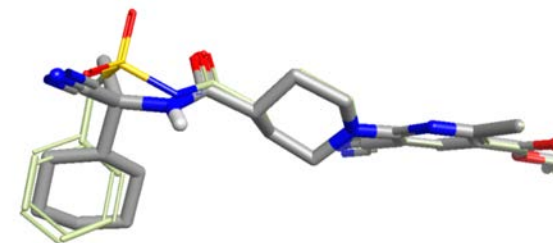
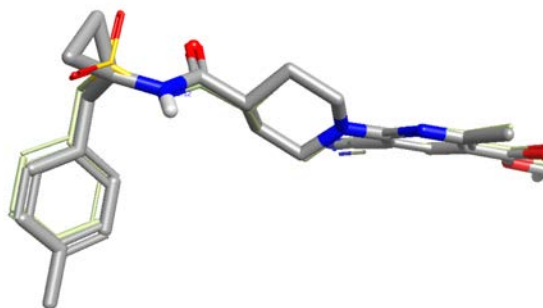
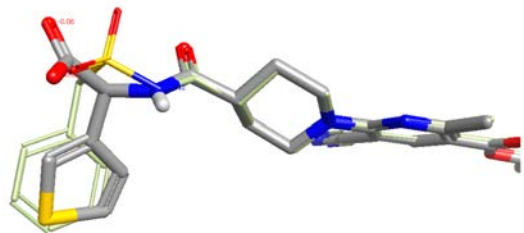
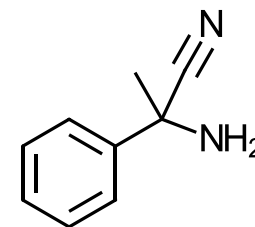
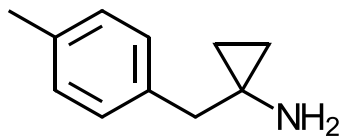
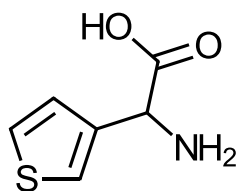
- > P2Y<sub>12</sub> is a GPCR chemoreceptor located on platelet cell membranes: inhibition of P2Y<sub>12</sub> is an important antithrombotic drug target.
- > Apply reagent databases to AZD1283



Opportunity to quickly explore SAR by using available amines. Non-acidic amides known to be active.

# P2Y<sub>12</sub> Selected Results

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# Conclusion

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- > Reagent based bioisostere databases good for
  - > finding imaginative new molecules
  - > Linking computational output to synthetic chemistry
- > ATPAT
  - > Great cheminformatics language
  - > Steep learning curve
- > Challenges
  - > Keeping reagent databases up to date
    - > KNIME/PP workflows

# Acknowledgements

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- > Mark Mackey
- > Martin Slater
- > Rae Lawrence



# Thank you

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tim@cresset-group.com

   cressetgroup

*Slides available on request*